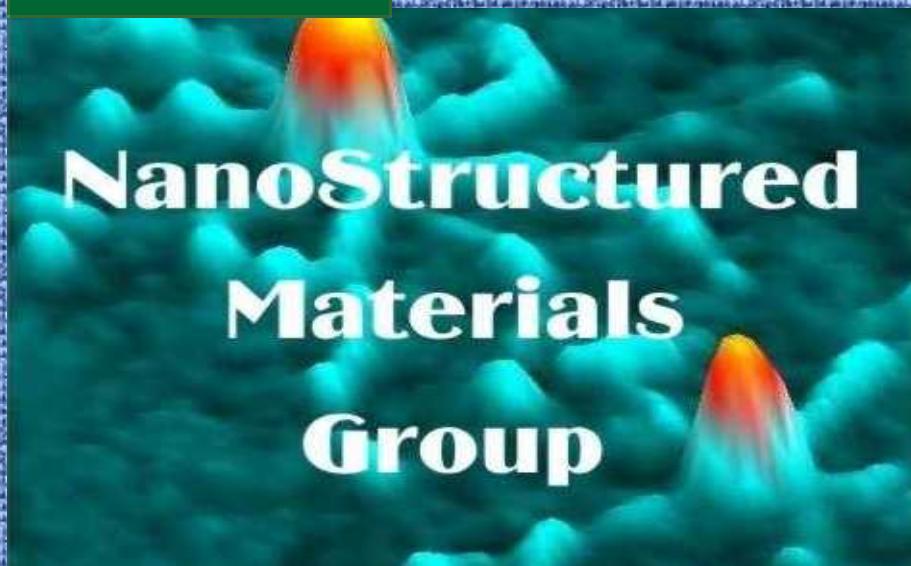


# Combinatorial approach to the problem of interfacial interactions via gradient polymer brushes

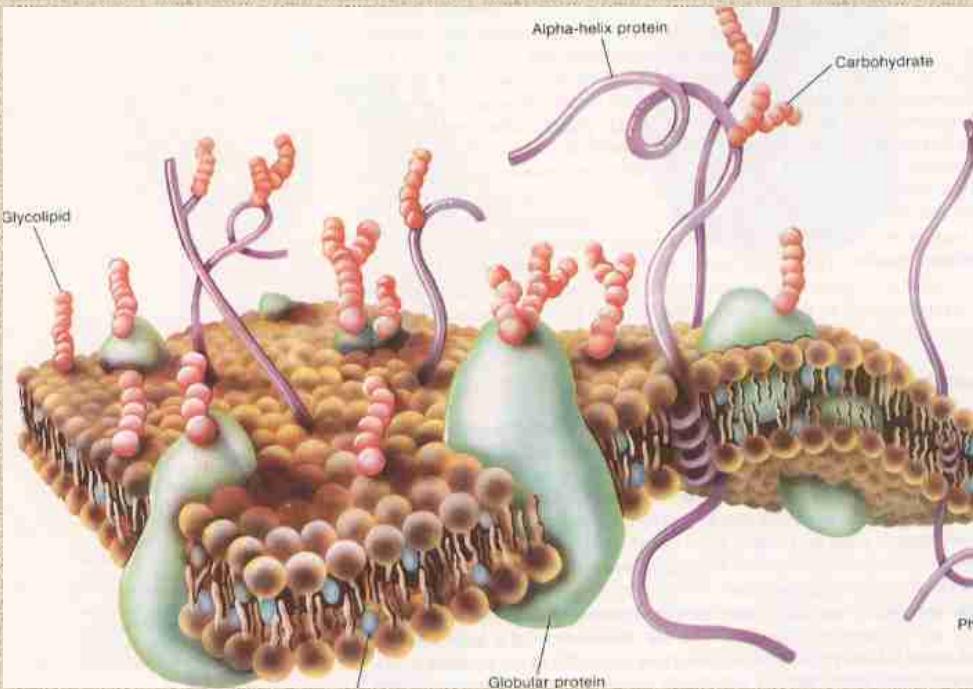
NCMC-9: Combinatorial Methods for Nanostructured Materials, April 24-25, 2006



**Sergiy Minko**  
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8 Clarkson Ave, Potsdam, New  
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(315) 268-2389 (voice),  
(315) 268-6610 (fax),  
e-mail: [sminko@clarkson.edu](mailto:sminko@clarkson.edu)

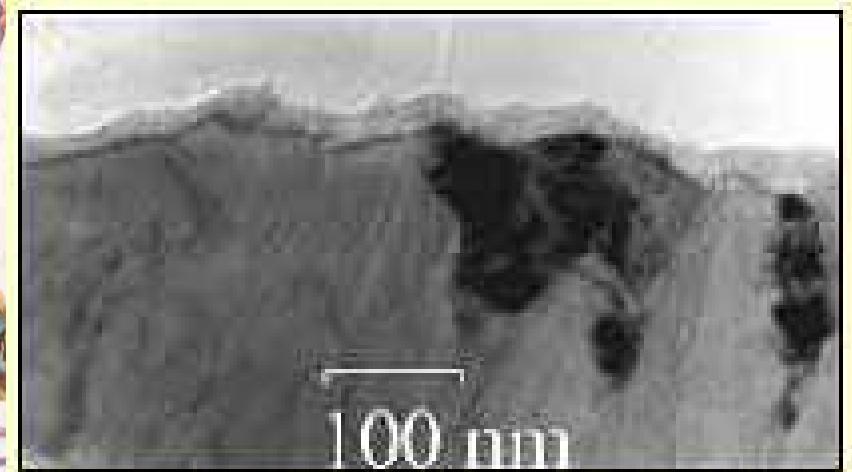
# Motivations:

Interactions between phases in complex systems play key role



Cell membrane

<http://www.hallym.ac.kr/~de1610/histology/cell-3.jpg>

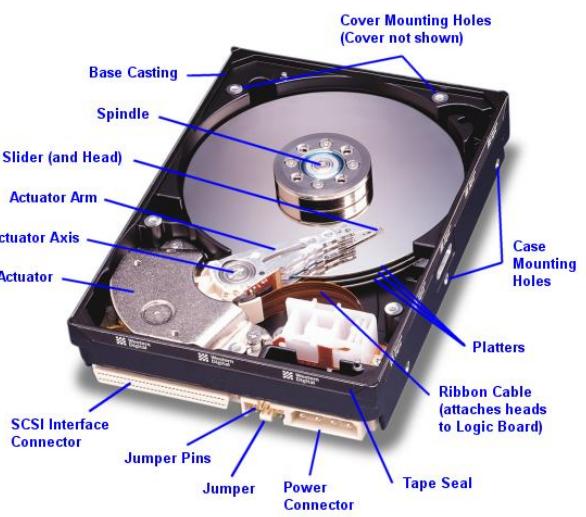


Epoxy adhesive on metal surface

<http://www.kratos.com/EApps/Adhapp2.html>

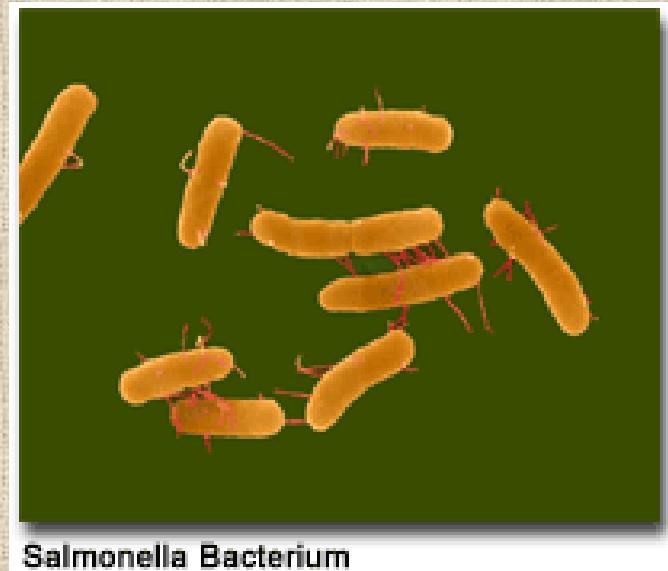
# Motivations:

Permanently strong interactions



Permanently weak interactions

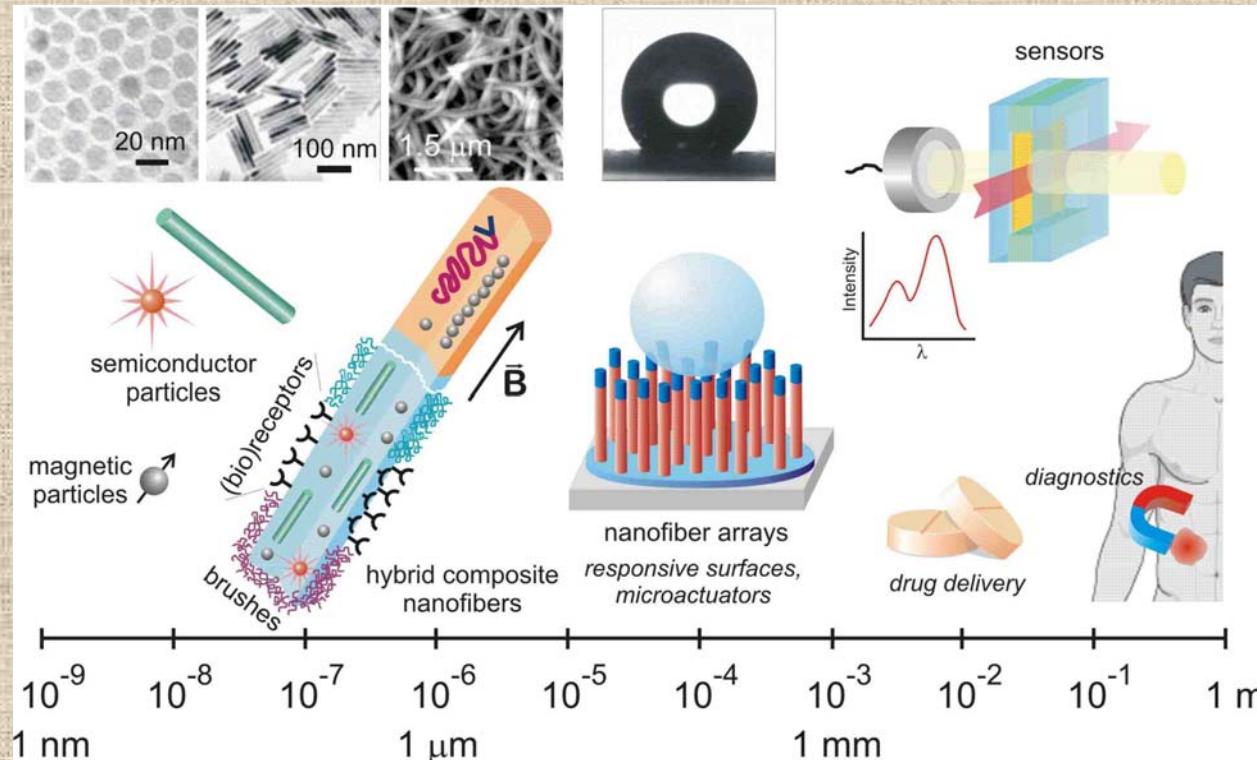
Highly selective, responsive,  
dynamically changeable interactions



**Salmonella Bacterium**

# Motivations:

- Precise regulation of interactions
- Responsive interactions
- Selective interactions

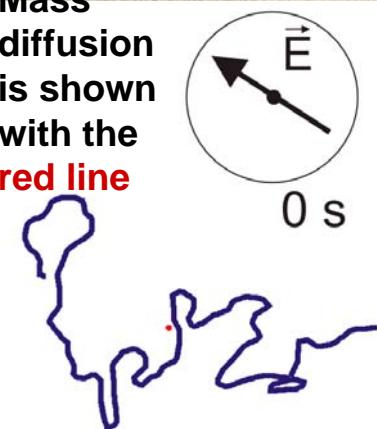


Hierarchically designed  
materials and devices perform  
based on precisely adjusted  
selective and responsive  
interactions

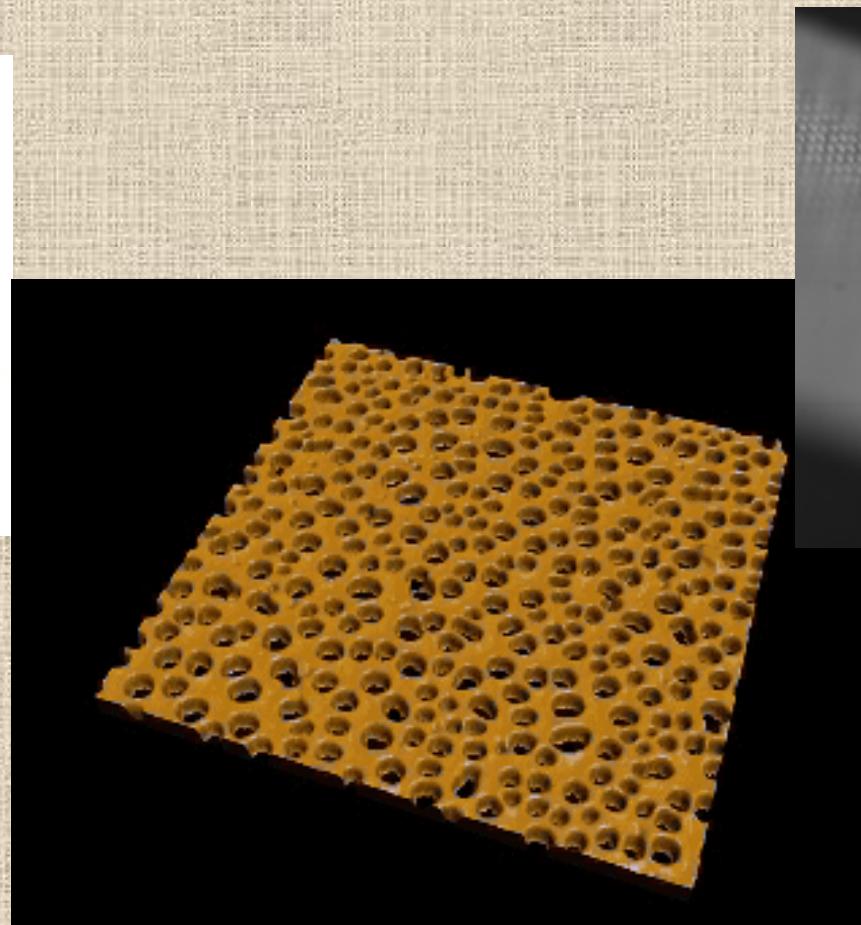
# Applications:

## Smart responsive materials and devices from drug delivery systems to construction materials

Center of  
Mass  
diffusion  
is shown  
with the  
**red line**



Diffusion of  
polyelectrolyte  
molecule in  
electrical field  
(NSF project)



Responsive polyelectrolyte membrane  
(current ARO project)

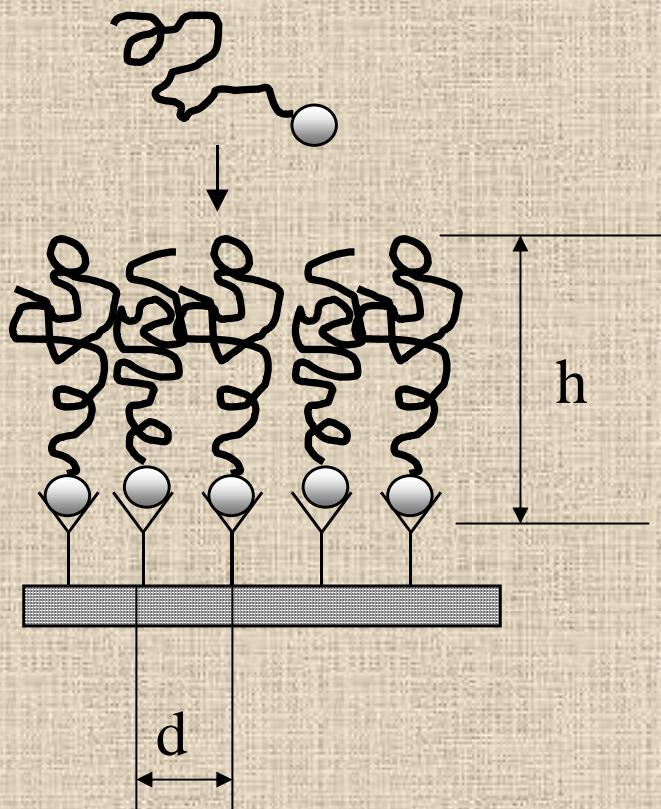
Ultrahydrophobic textile  
(current NTC project with  
Igor Luzinov, Clemson  
University)

# Contents:

- What is polymer brush?
- Why polymer brushes?
- Properties
- Complex polymer brushes
- Gradient polymer brushes
- Examples

# What is polymer brush?

End-tethered polymer chains



$$\Delta f = f_1 + f_2.$$

$$f_1 \approx kT h^2 R_g^{-2};$$

where  $R_g \approx N^{1/2}a$  (*chain dimension*),

*a-size of the monomer,*

$$f_2 \approx kT v N(N\sigma/h),$$

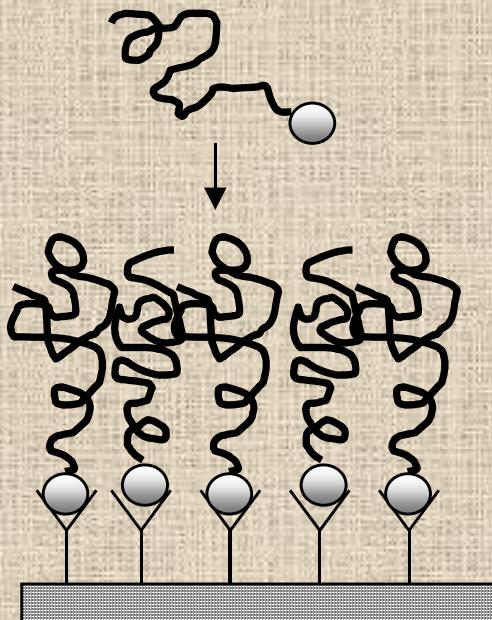
where  $\sigma = 1/d^2$  (*grafting density*)

$$h \approx N(v\sigma a^2)^{1/3};$$

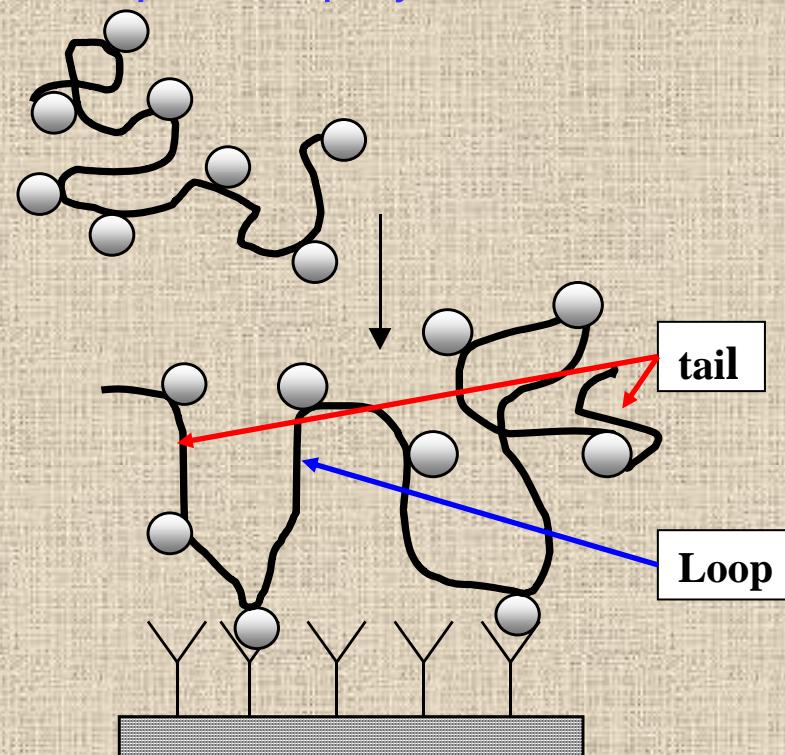
$$f = kTN(v\sigma a^{-1})^{2/3}.$$

# Why polymer brushes?

1. Ability to regulate interaction due to the stretching of the polymer chains – not possible with low molecular weight surfactants.
2. Well defined architecture of the interfacial layers – not the case for adsorbed, randomly grafted, or LBL deposited polymers



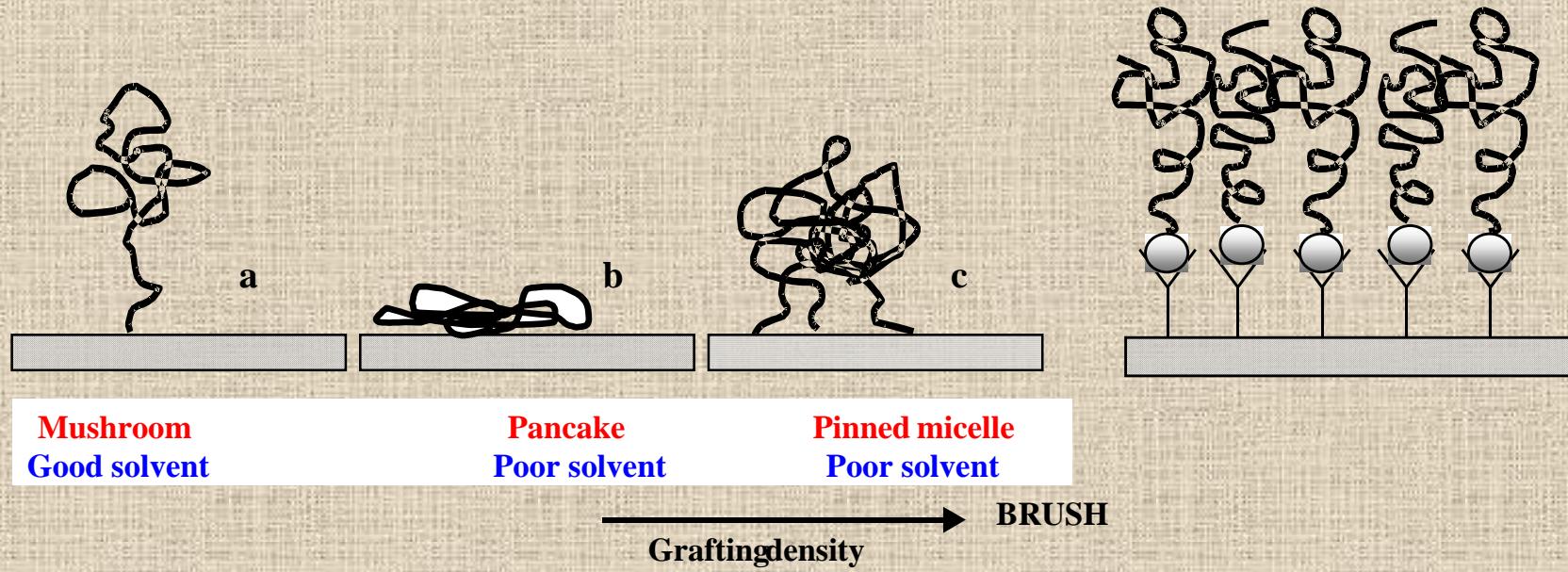
Schematic illustration of end tethered polymer chains forming brush-like layer - well defined architecture



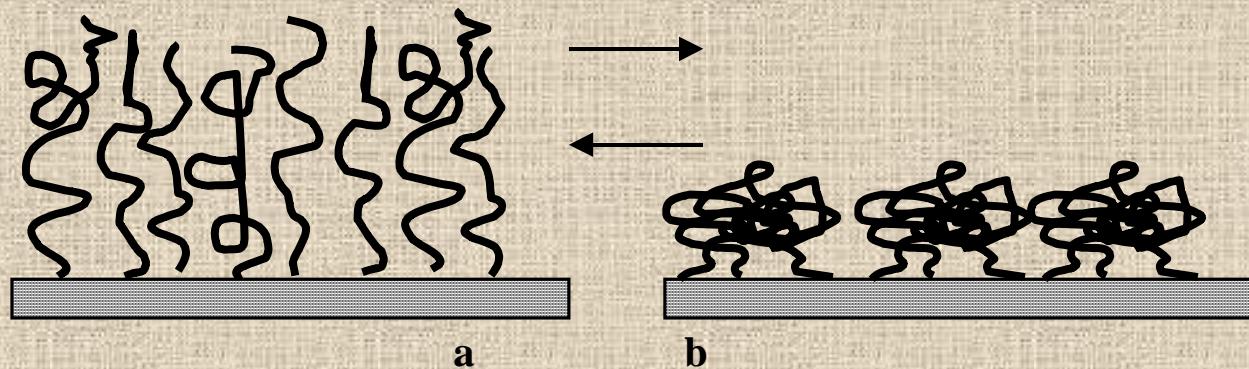
Schematic illustration of the polymer chain grafted via side functional groups. Loops and tails behave similar to polymer brush – less defined architecture

# PROPERTIES: interaction with solvent

Different concentration regimes for tethered polymers

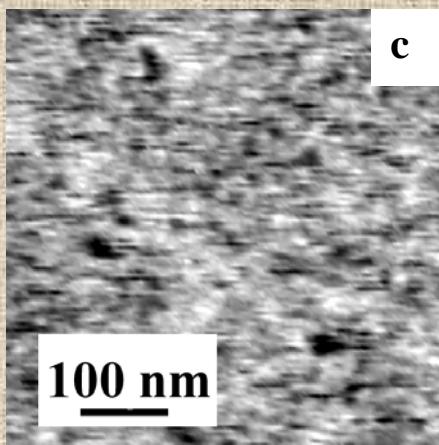


# PROPERTIES: interaction with solvent

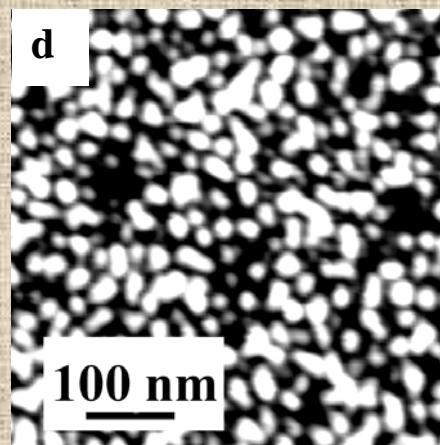


Good solvent

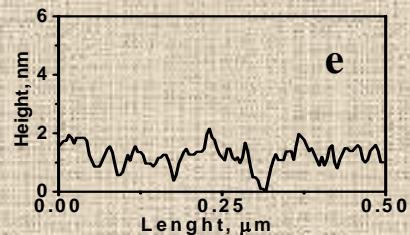
Poor solvent



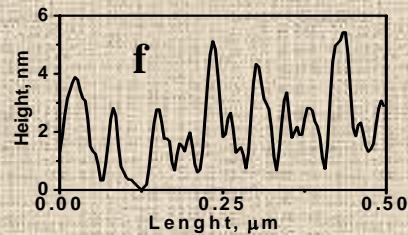
100 nm



100 nm

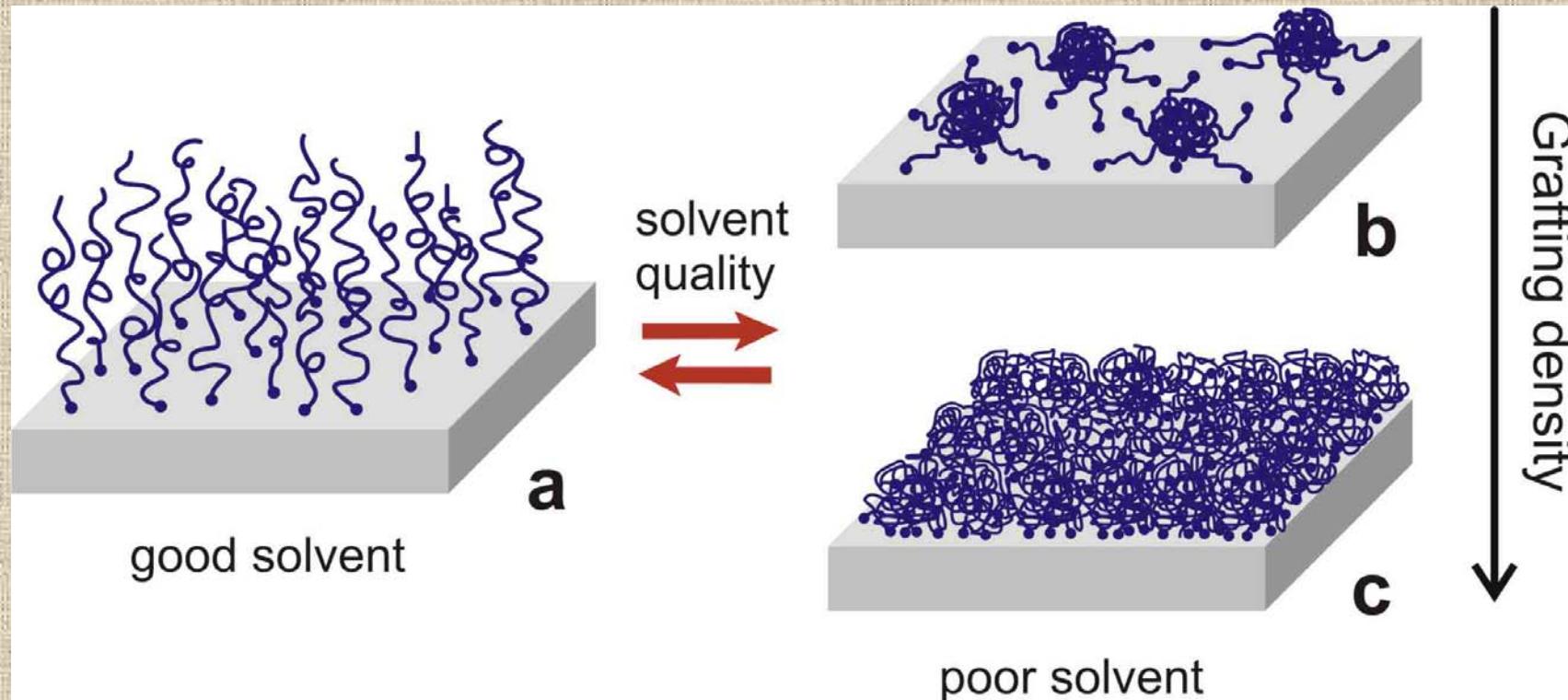


e



f

# PROPERTIES: interaction with solvent



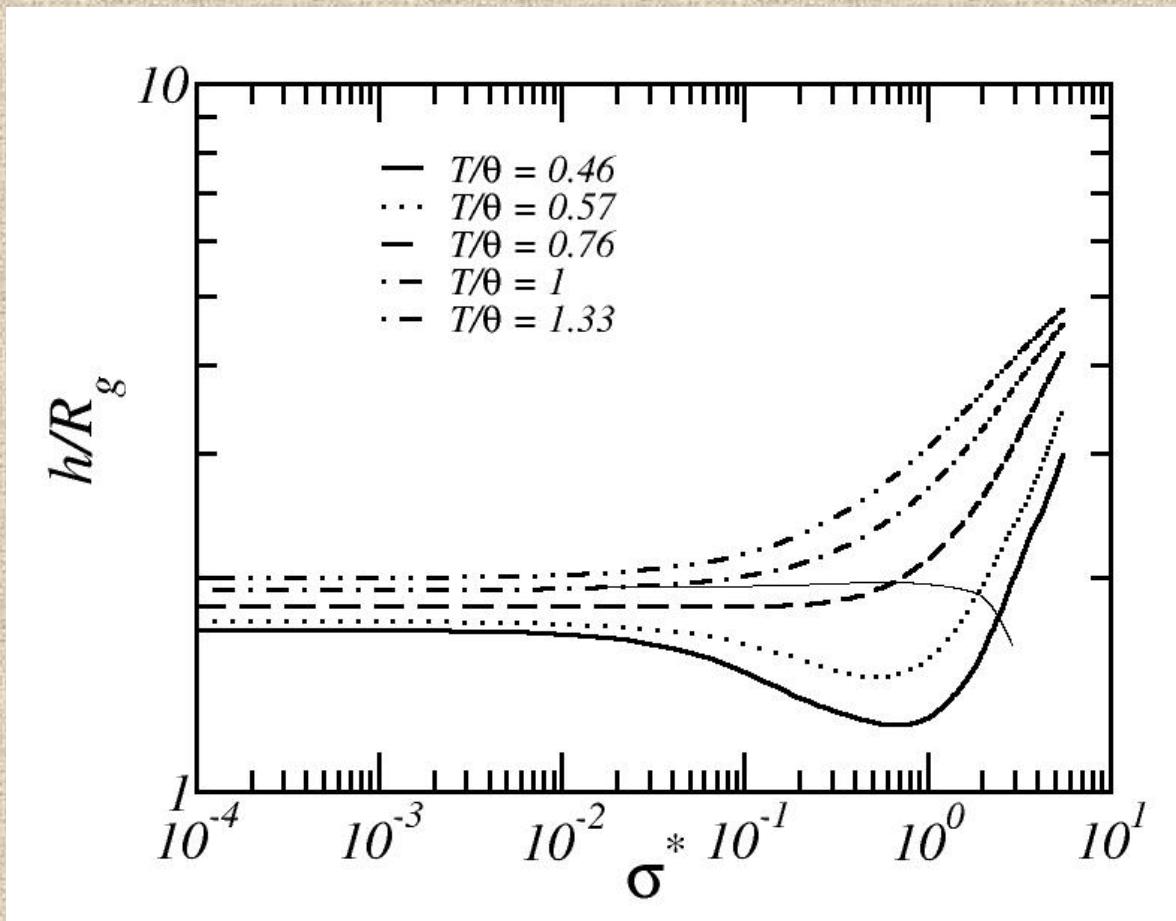
**Polymer brush properties as function of solvent quality and grafting density**

A – stretched brush regime

B- pined micelles regime

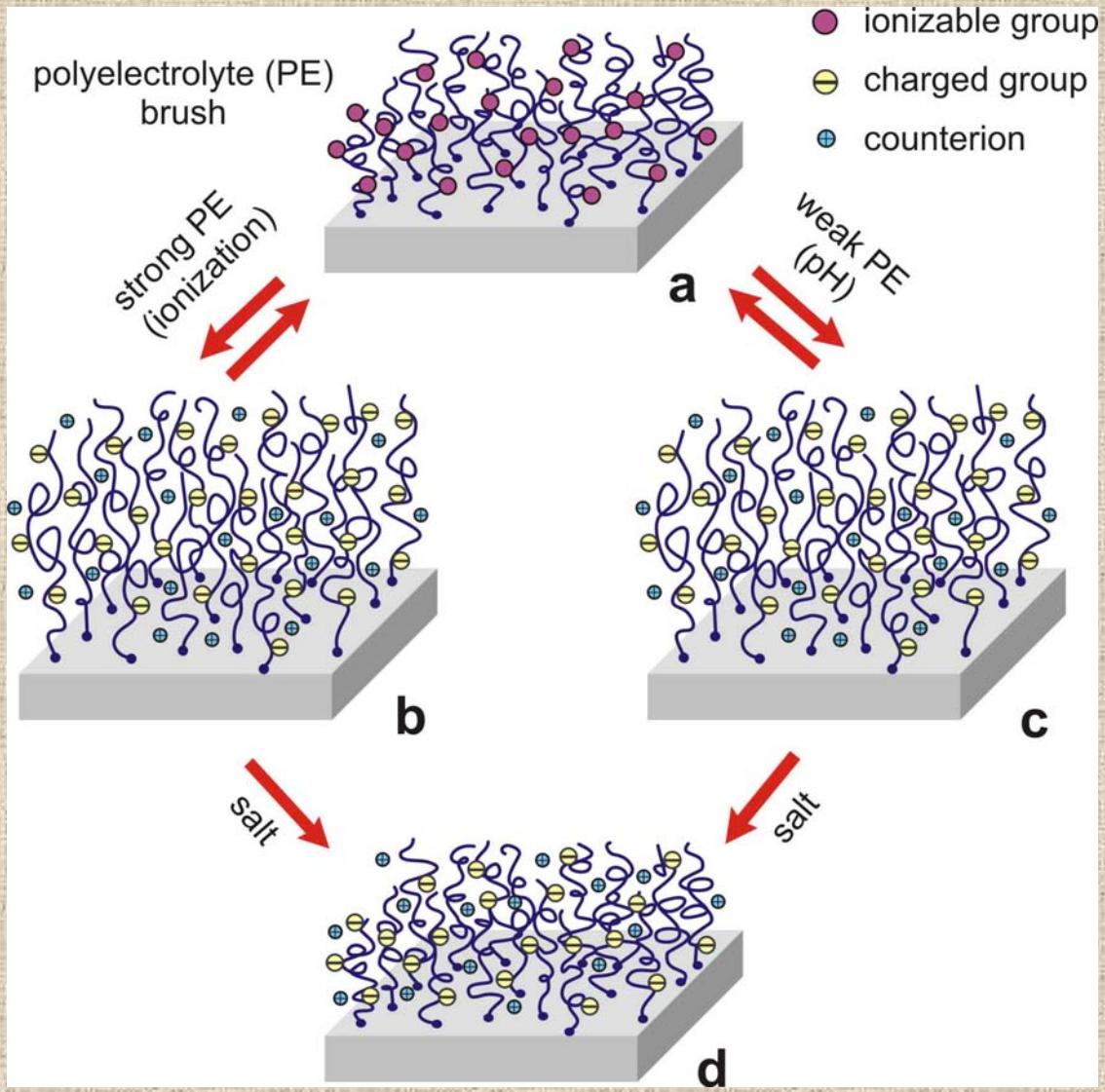
C- collapsed brush regime

# PROPERTIES: interaction with solvent

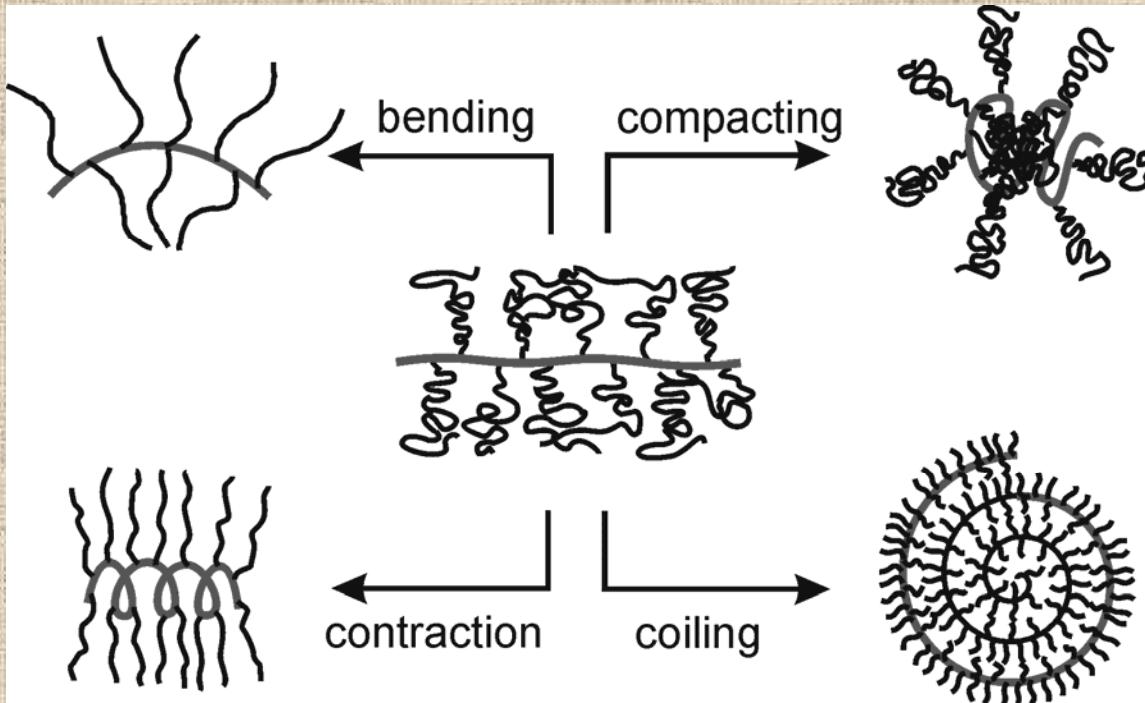


**Height of the brush layer, scaled by the bulk radius of gyration of the chain in good solvent, as a function of the scaled surface coverage for a variety of qualities of solvents.** Szleifer, I.; Carignano, M. A. "Tethered polymer layers", *Adv. Chem. Phys.* 1996, 94, 165-260.

# Diversity of polymer brushes: polyelectrolyte brushes



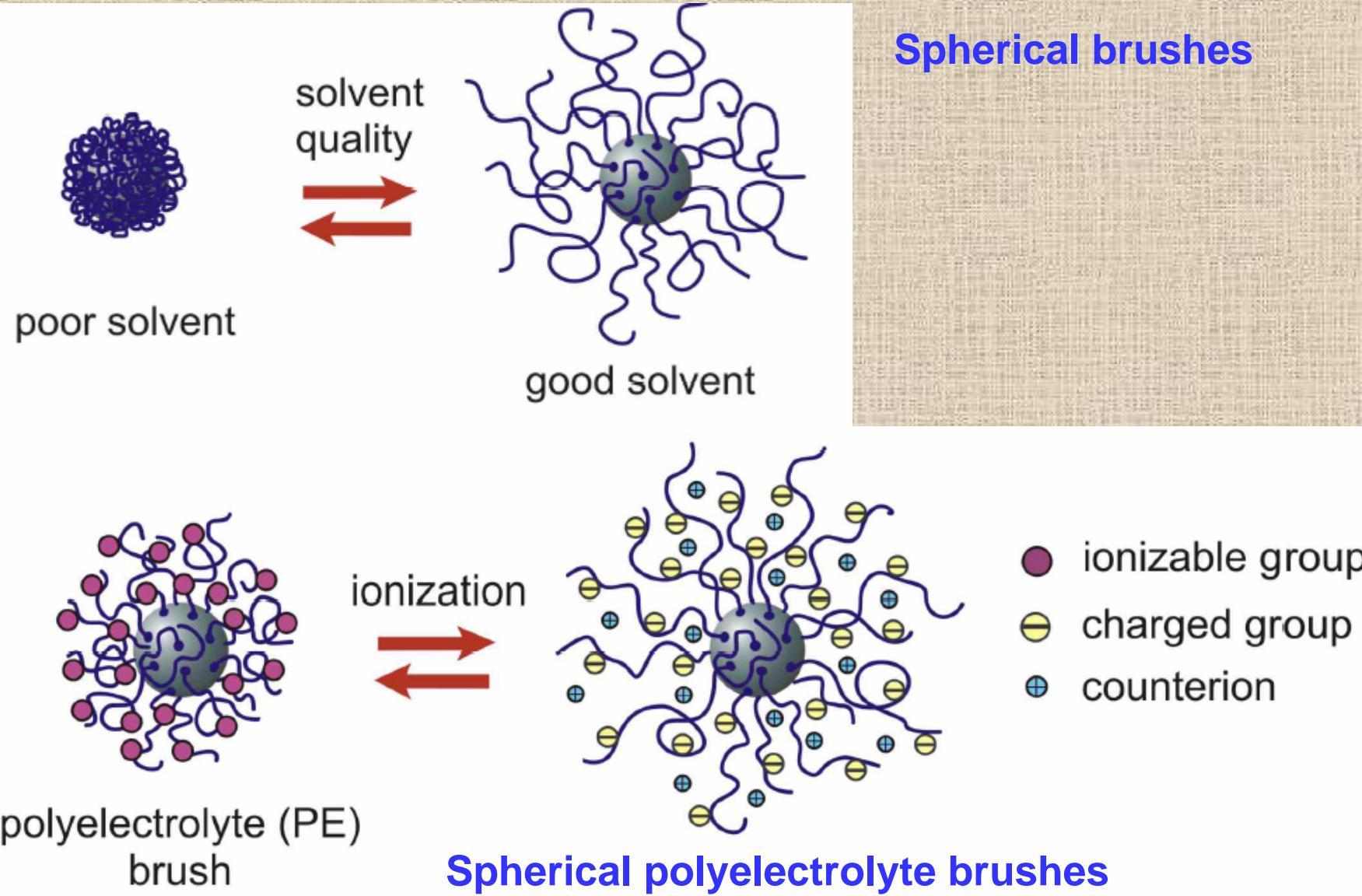
# Diversity of polymer brushes:



## Molecular brushes

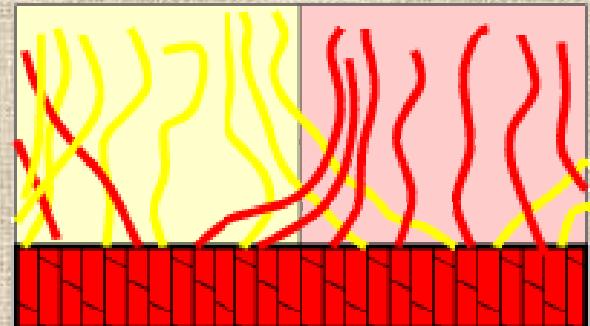
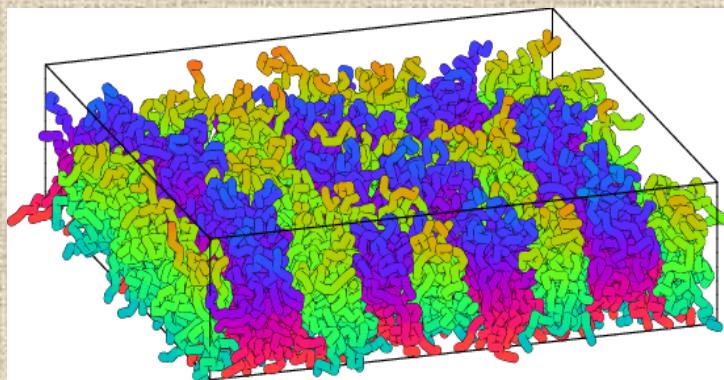
Boyce, J. R.; Sun, F. C.; Sheiko, S. "Stimuli-responsive brush-like macromolecules", In *Responsive polymer materials: design and applications*; Minko, S., Ed.; Blackwell Publishing Professional: Ames, 2006; pp 1-21

# Diversity of polymer brushes:



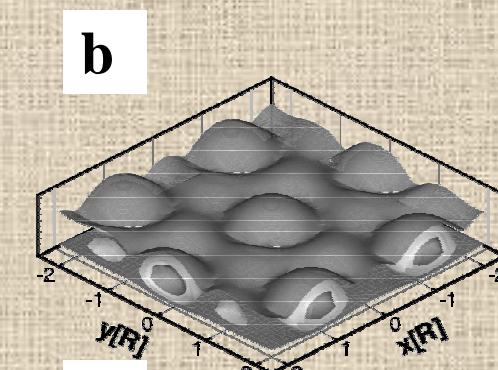
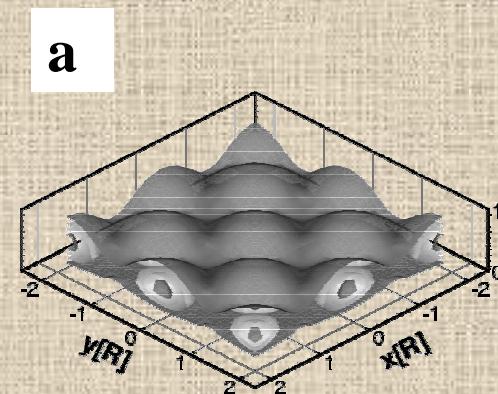
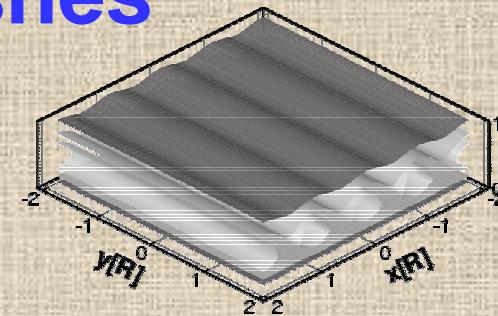
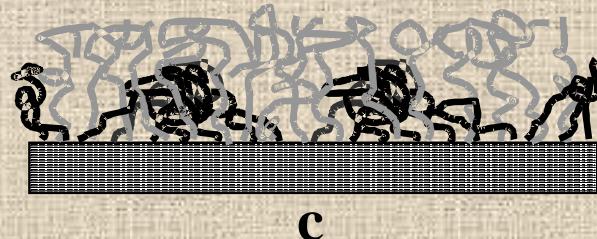
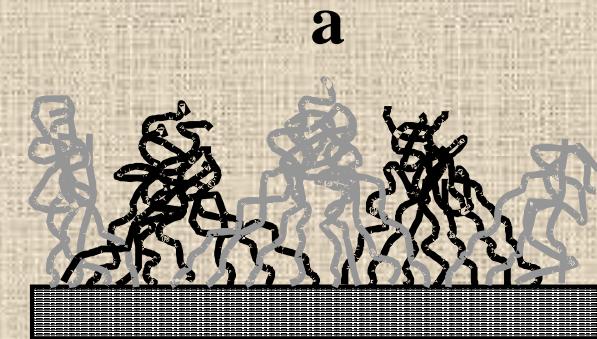
# Complex polymer brushes: Mixed polymer brushes

brush of two incompatible  
polymers in a common solvent

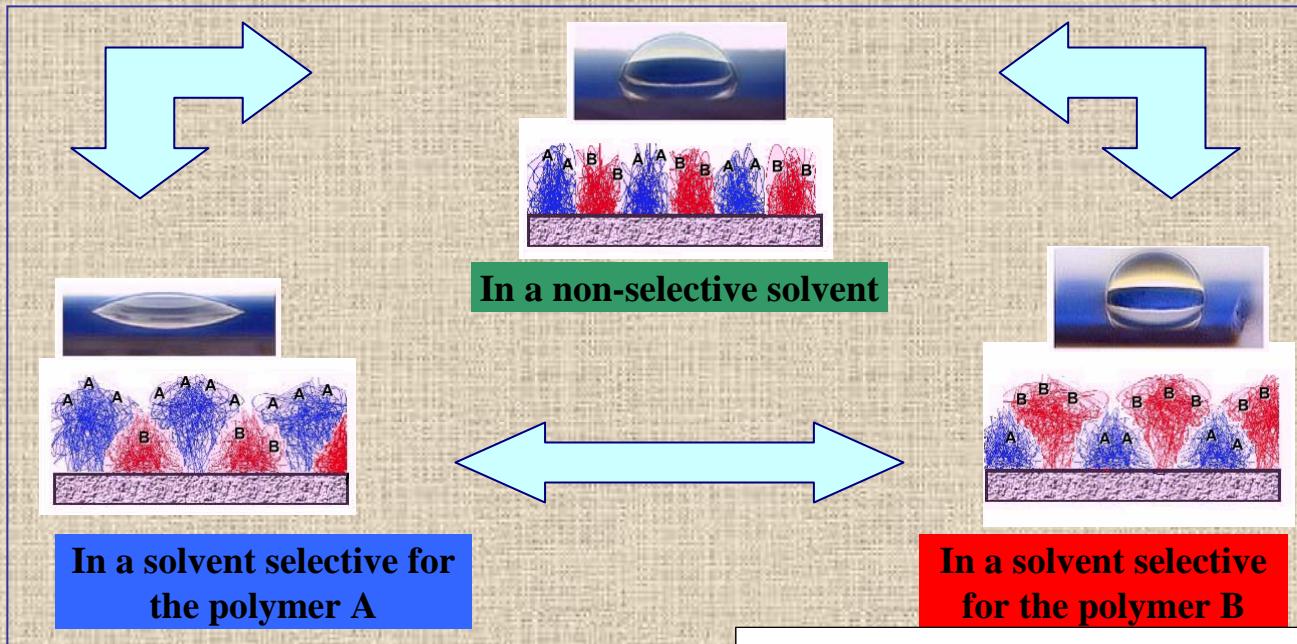


chains avoid energetically unfavorable  
contacts and pay entropic penalty

# Complex polymer brushes: Mixed polymer brushes

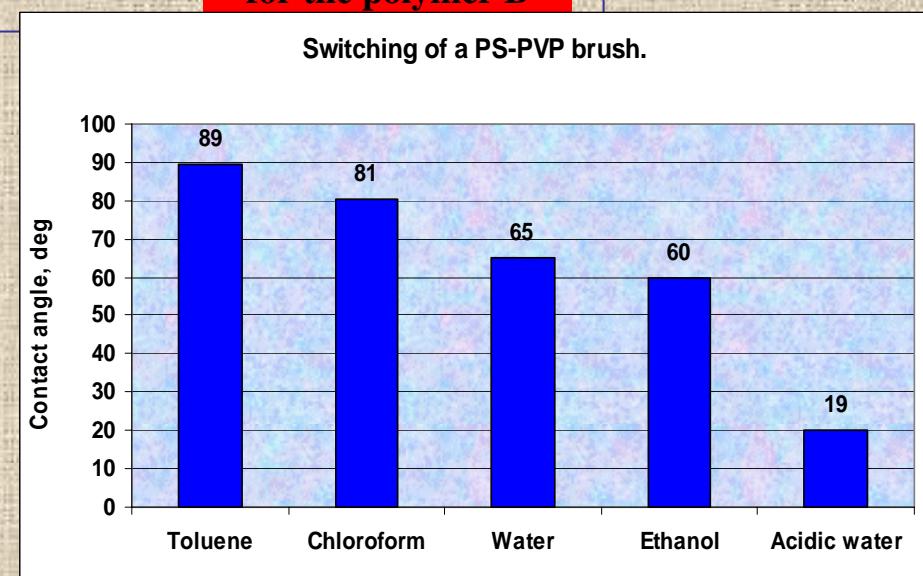


# Switching/adaptive properties of mixed polymer brushes

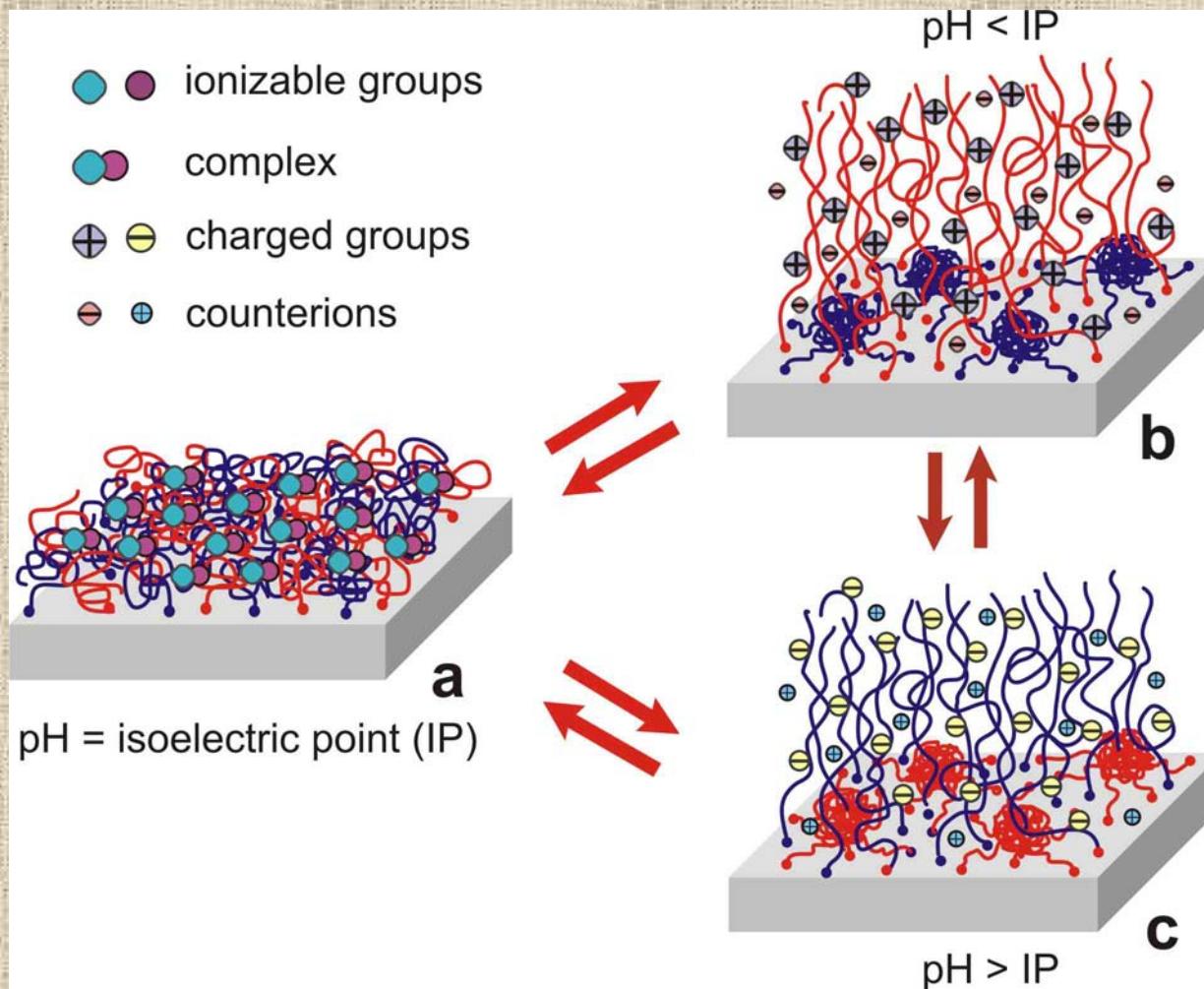


*Langmuir, 2002,  
18, 289-296*

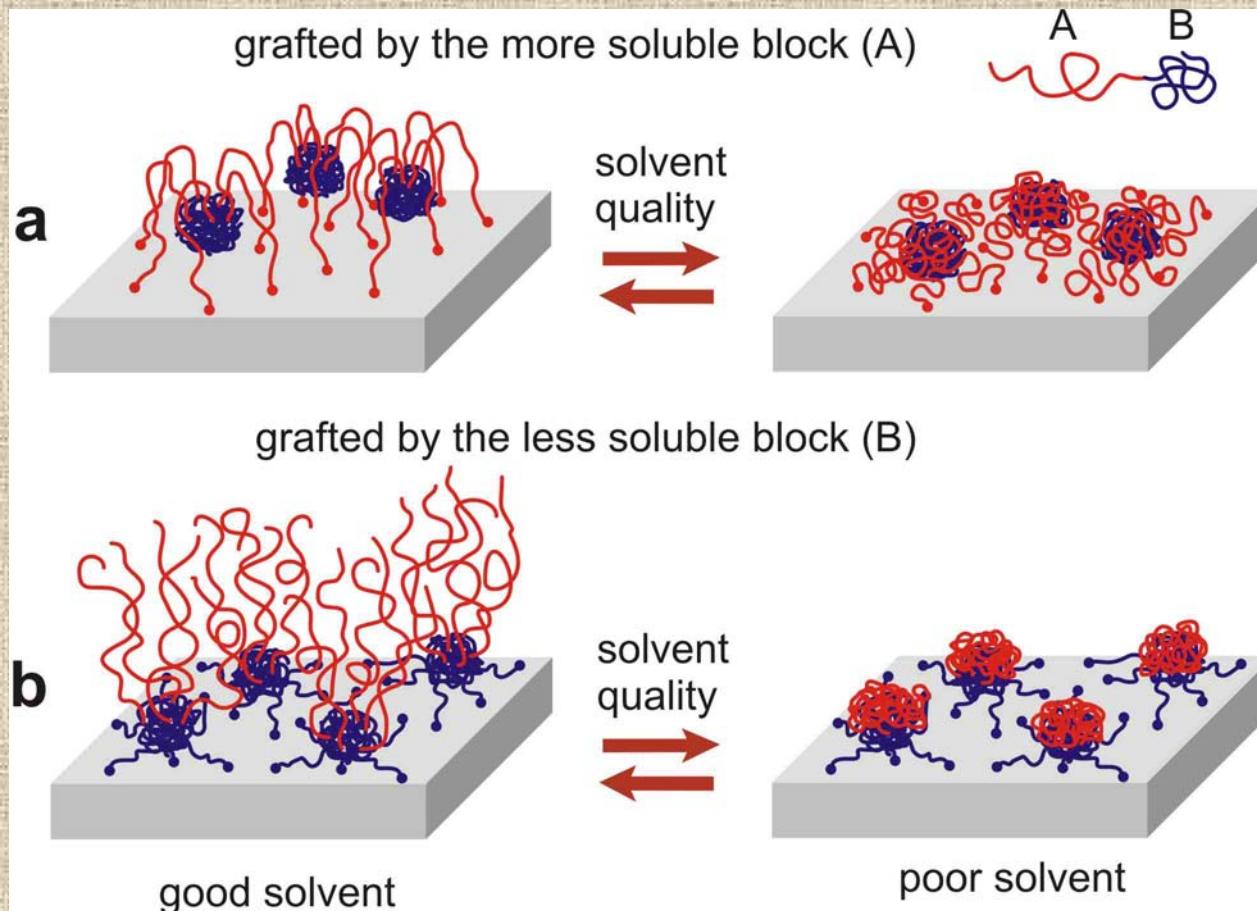
*Macromol. Rapid. Commun.,  
2001, 22, 206-211*



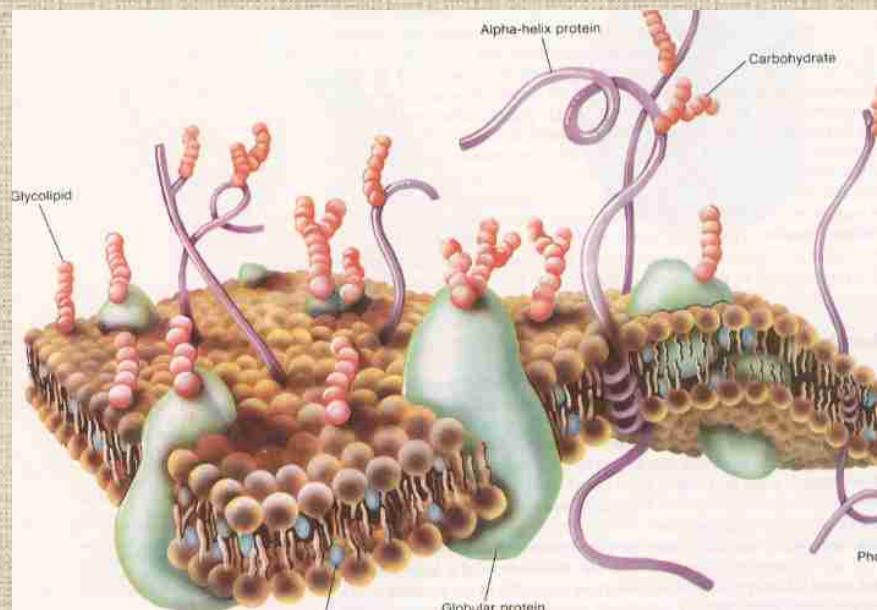
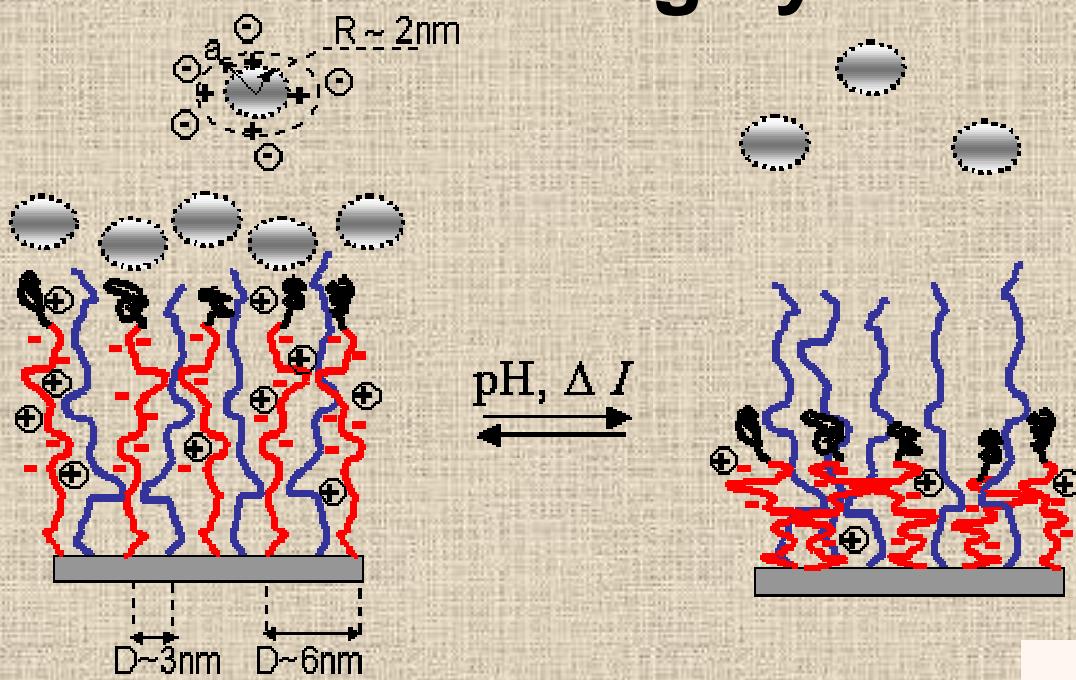
# Complex polymer brushes: Mixed polyelectrolyte brushes



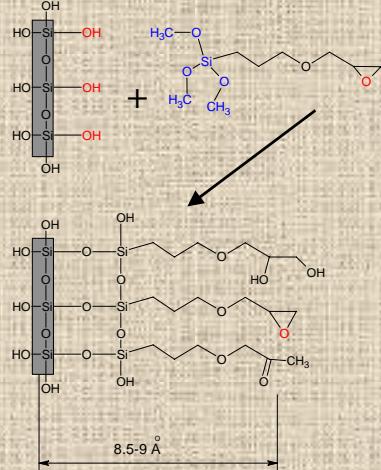
# Complex polymer brushes: Block-copolymer brushes



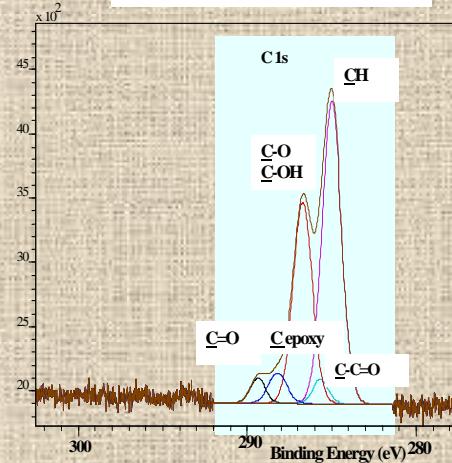
# Complex interactions – mimicking living systems



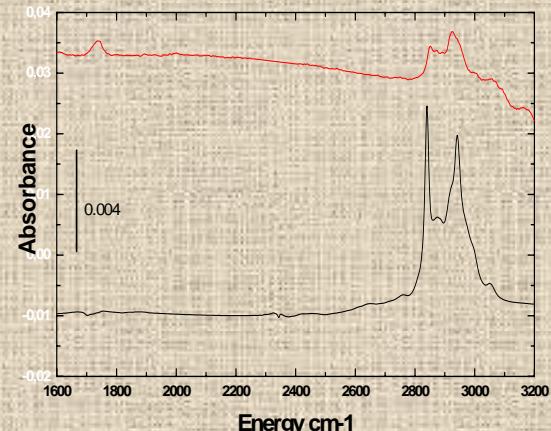
# SYNTHESIS



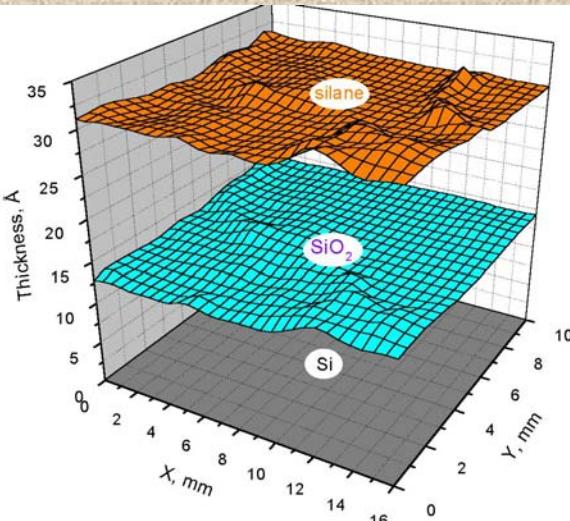
XPS of epoxy-silane on Si-wafer



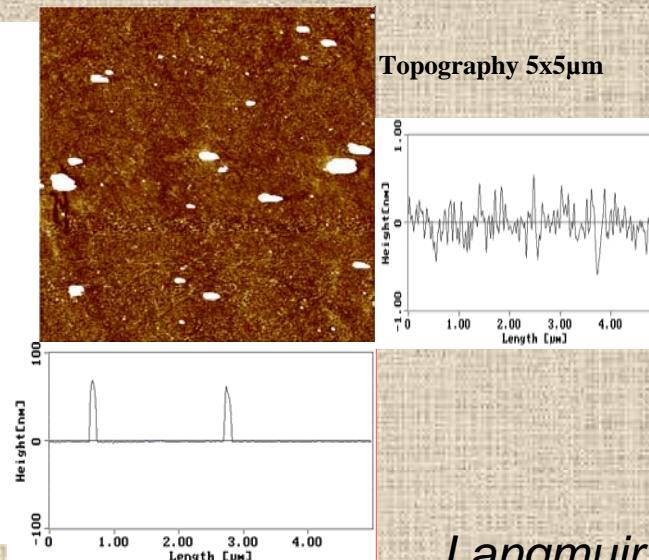
FTIR-ATR of epoxy-silane on Si-wafer



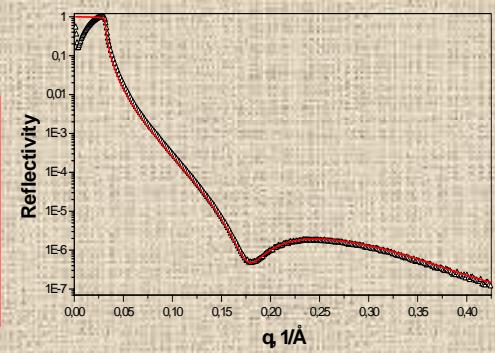
Ellipsometric topography of the epoxy-silane layer



AFM-imaging of the epoxy-silane layer

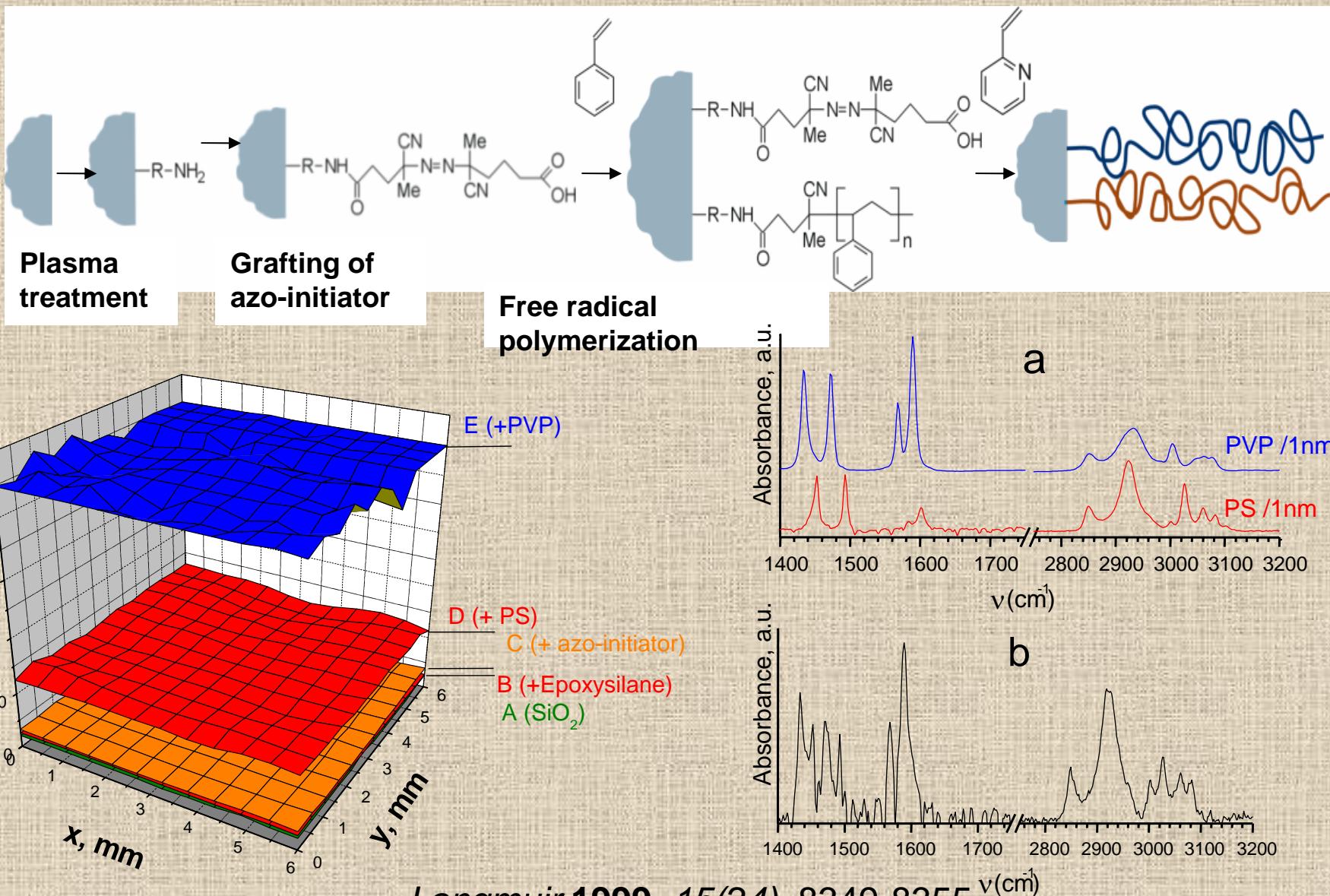


X-ray reflectivity



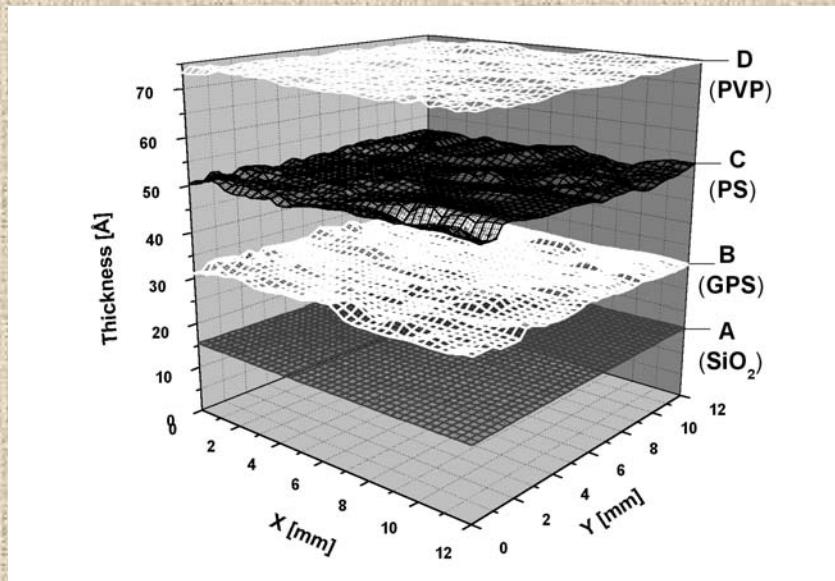
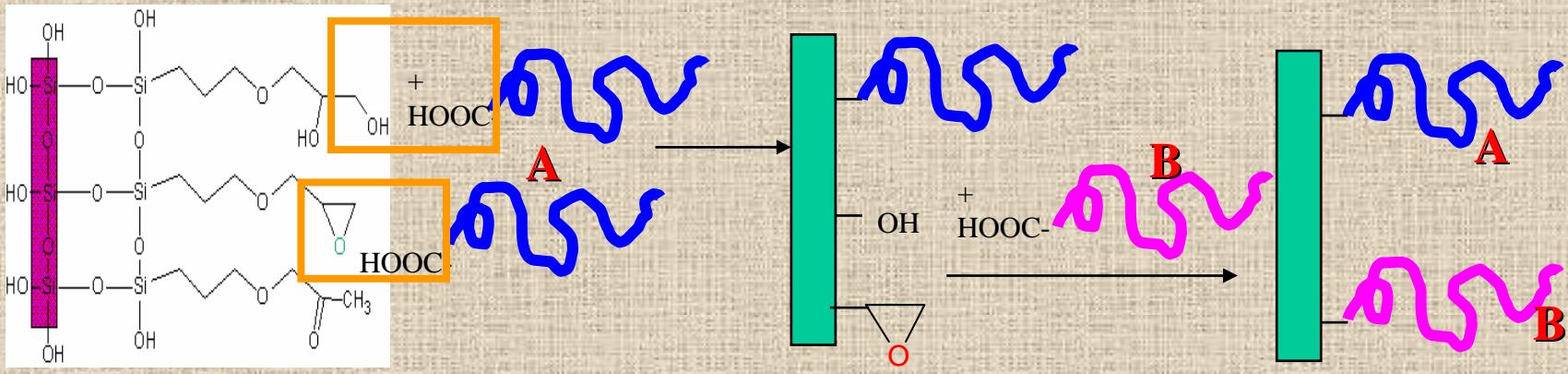
Epoxy-silane  
h = 1.7 nm, RMS = 0.42 nm

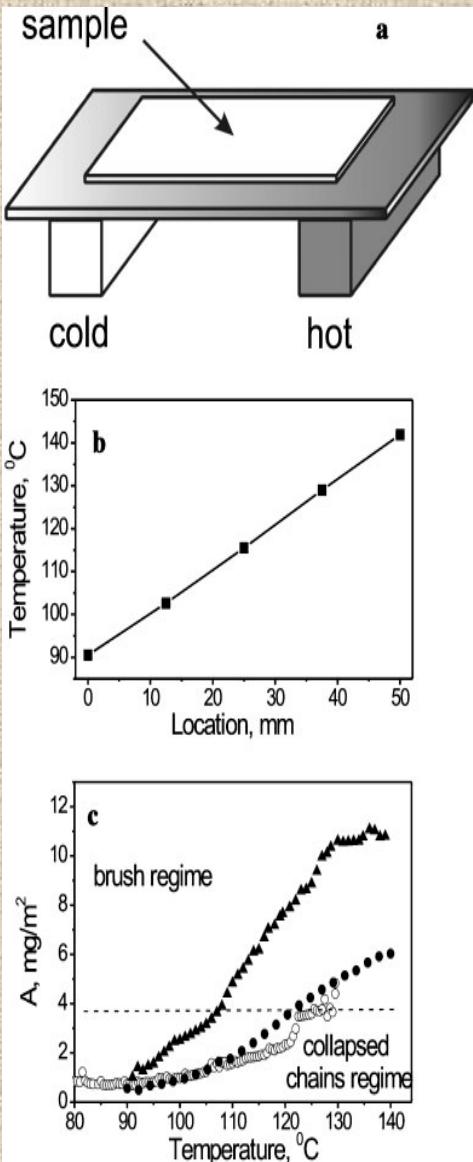
# SYNTHESIS: grafting from



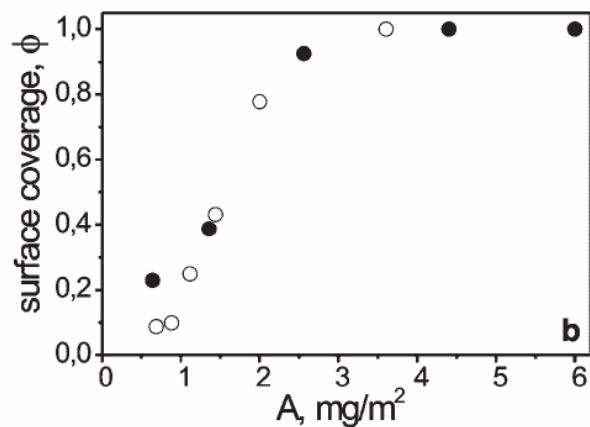
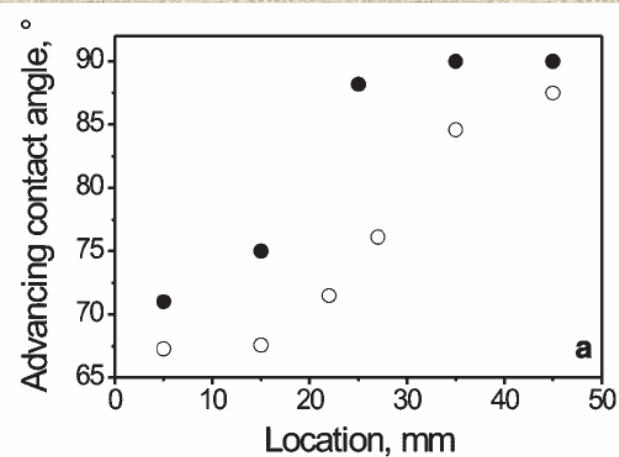
Langmuir 1999, 15(24), 8349-8355

# SYNTHESIS: grafting to



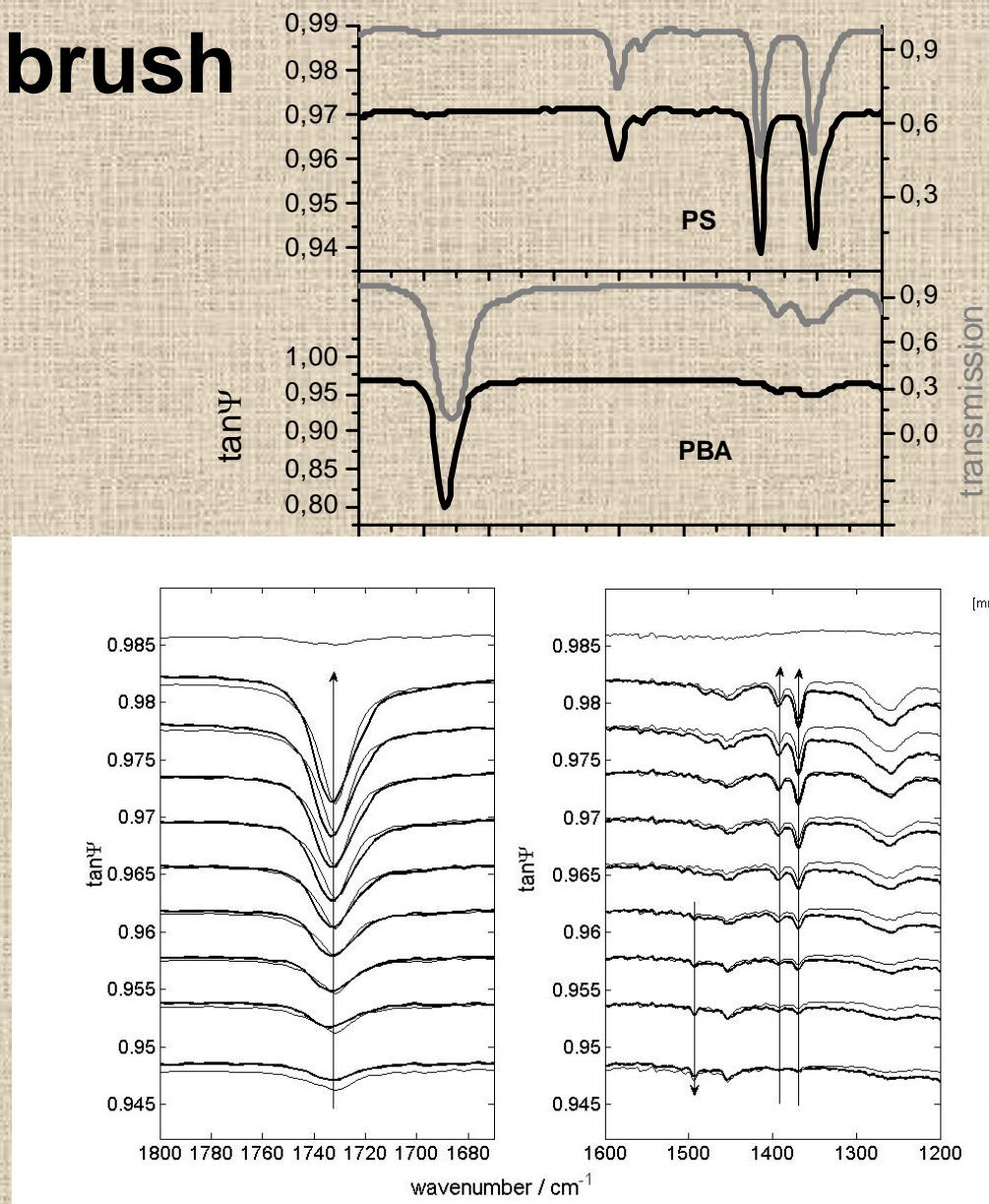
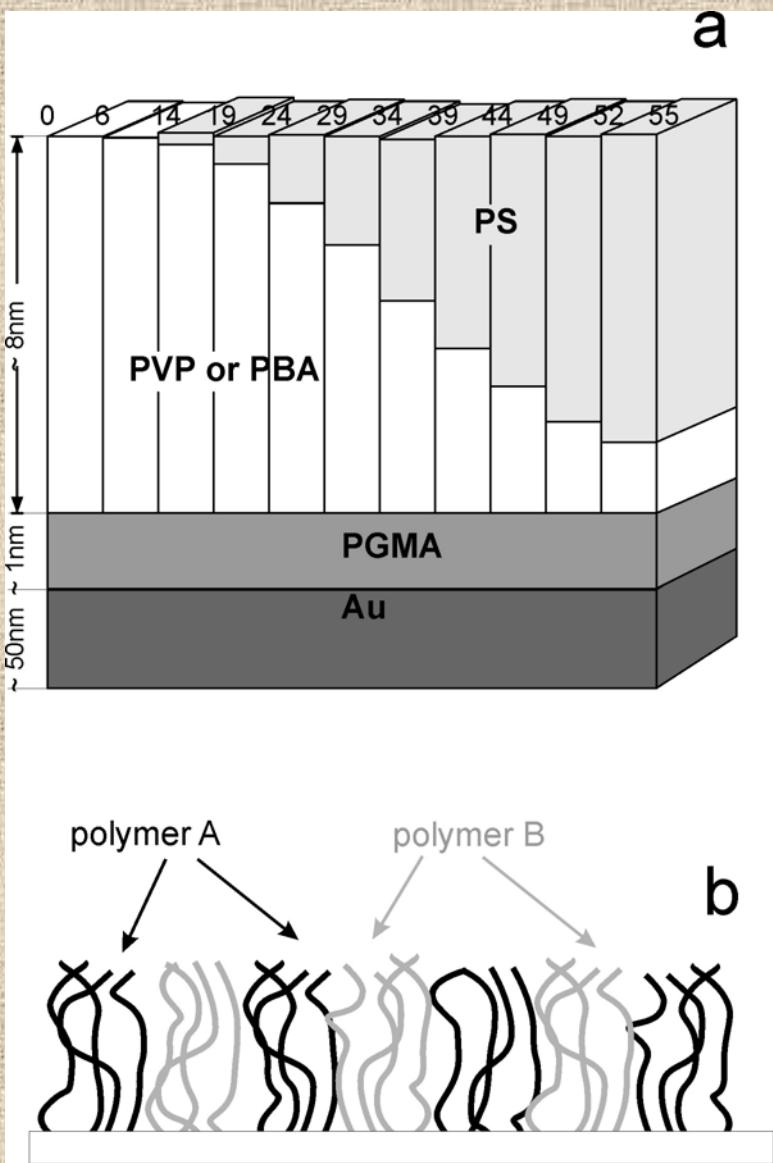


# Gradient Polymer Brushes by “Grafting To” Approach

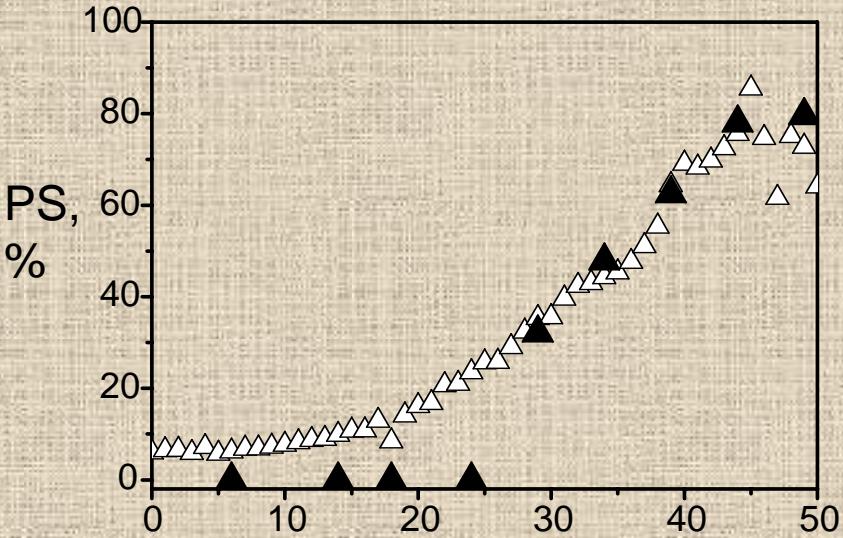
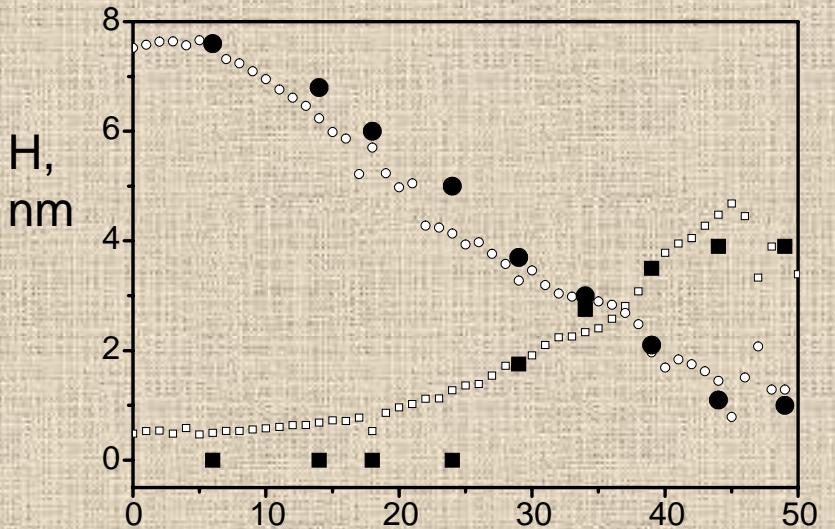
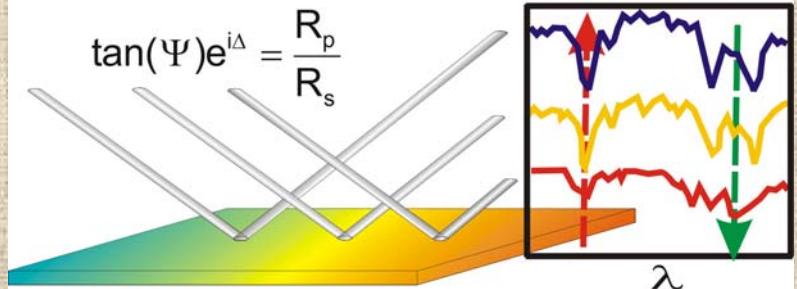


Macromol. Rapid Commun. 2004, 25, 360–365

# Gradient mixed brush



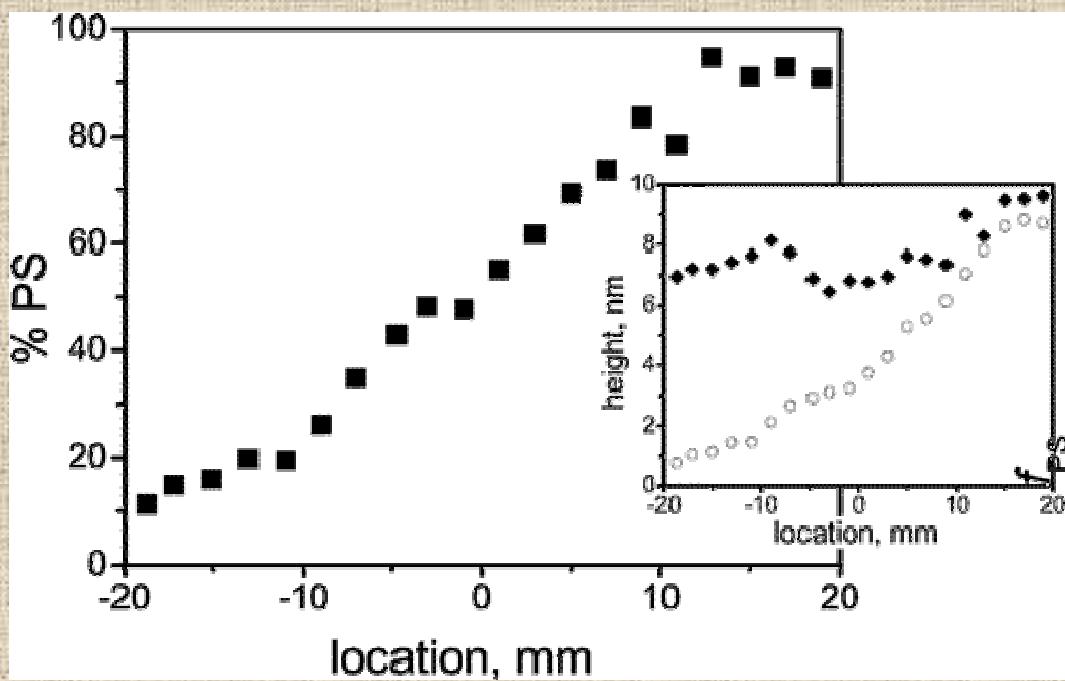
# Gradient mixed brush



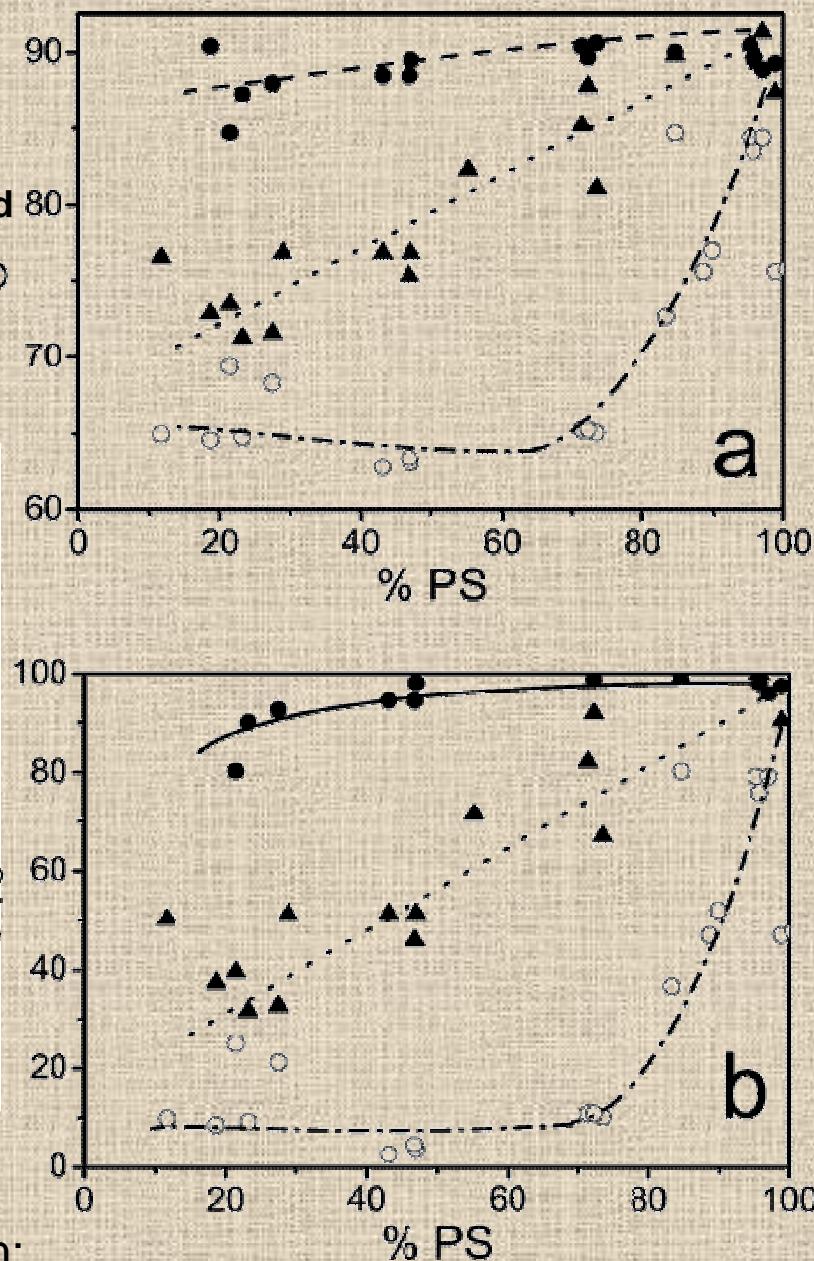
Ellipsometric mapping of thickness (a) and composition (b) of gradient PS-*mix*-PBA brush from single wavelength ellipsometry and IR-ellipsometry (solid)

# Switching gradient

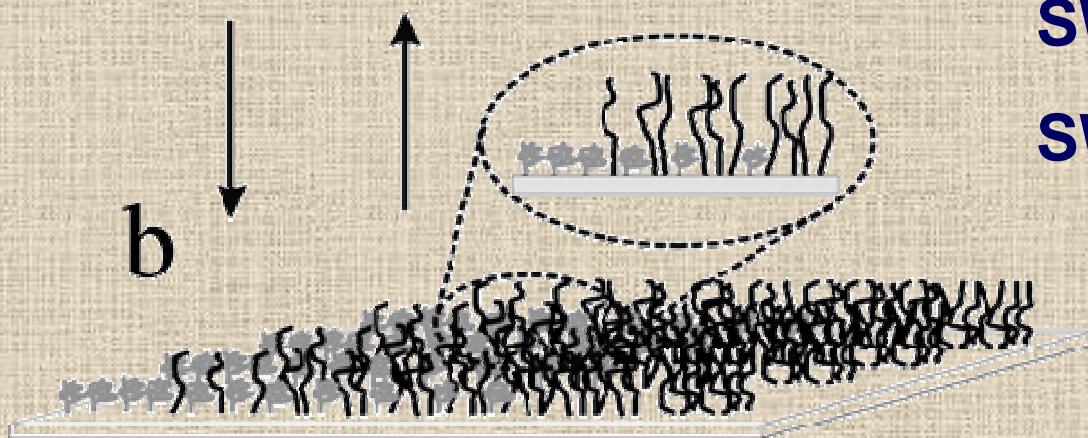
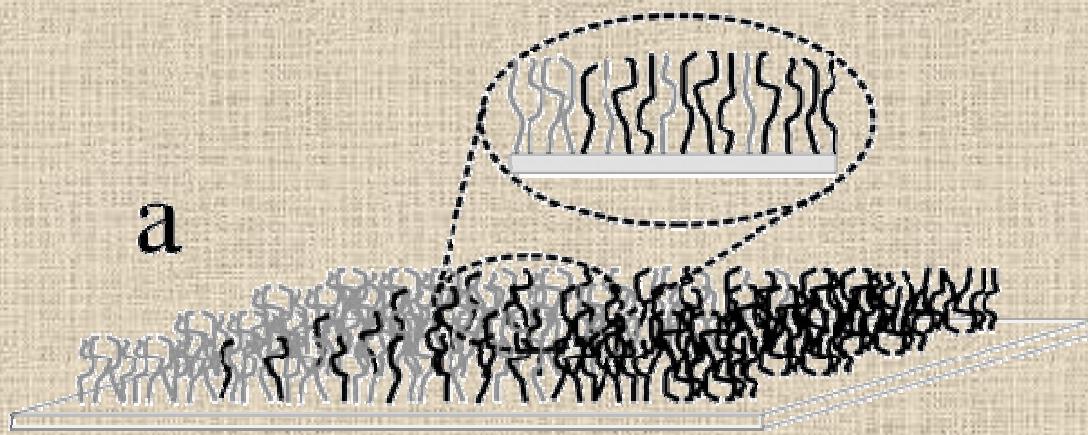
Fraction of grafted polystyrene vs location of the measured spot (on X axis) for the gradient PS-*mix*-P2VP brush. The inset demonstrates the ellipsometric thickness of the homopolymer PS and PS-*mix*-P2VP brushes (empty circles, PS; black circles, PS-*mix*-P2VP).



black circles, toluene; black triangles, chloroform;  
empty circles, ethanol

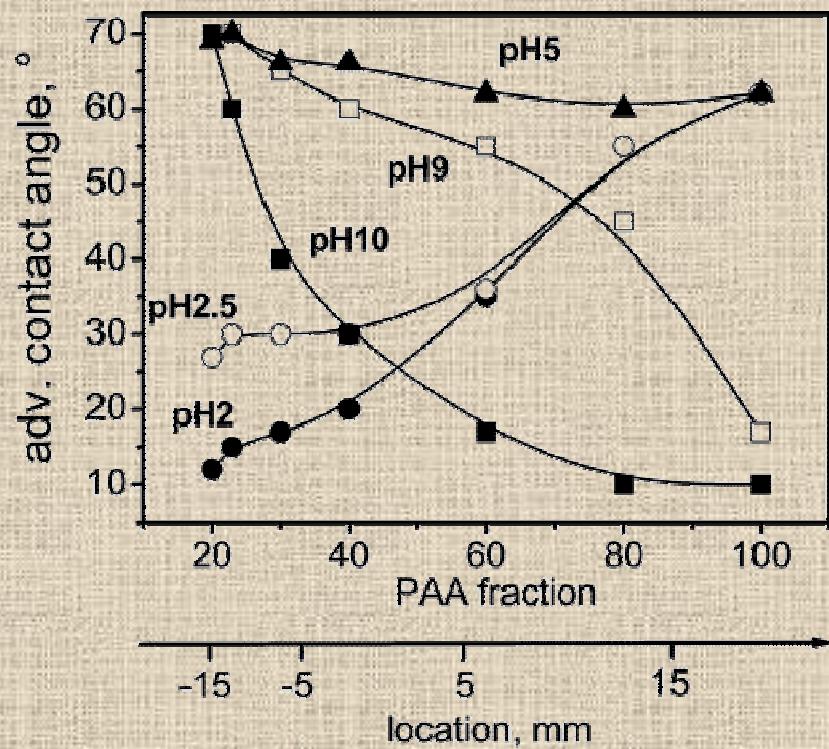
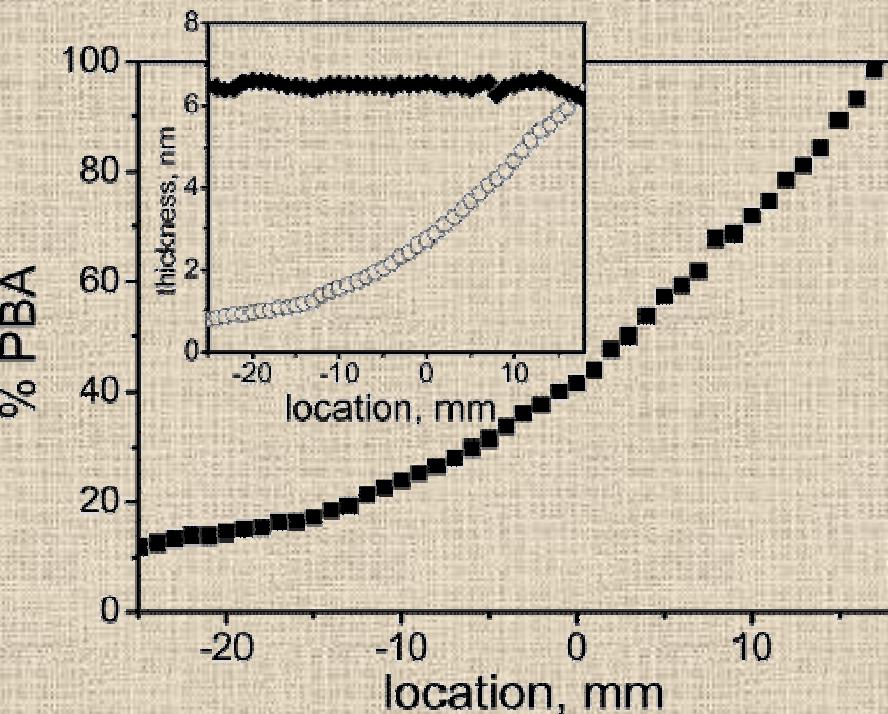


## GRADIENT:

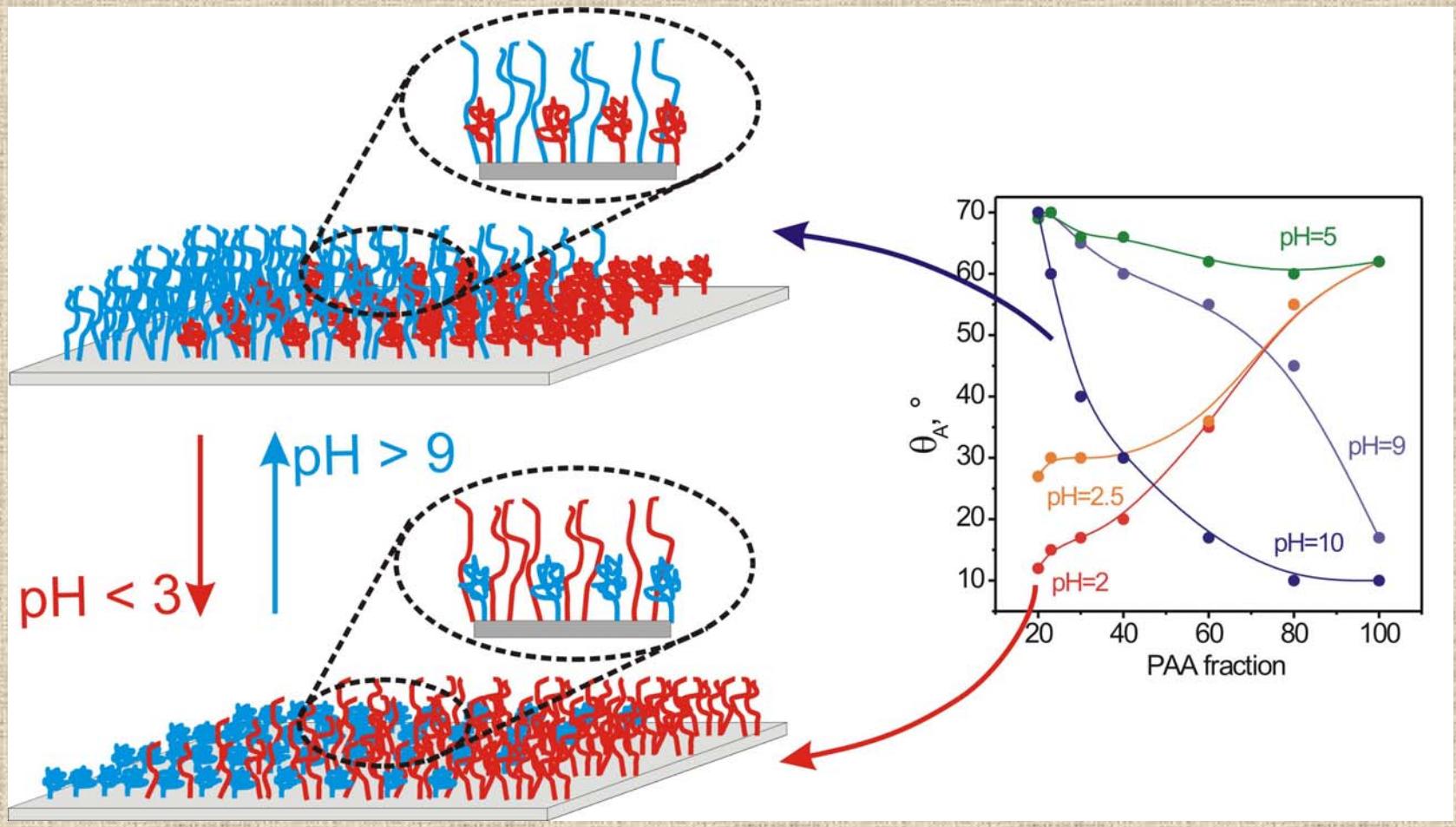


**SWITCHING ON  
SWITCHING OFF**

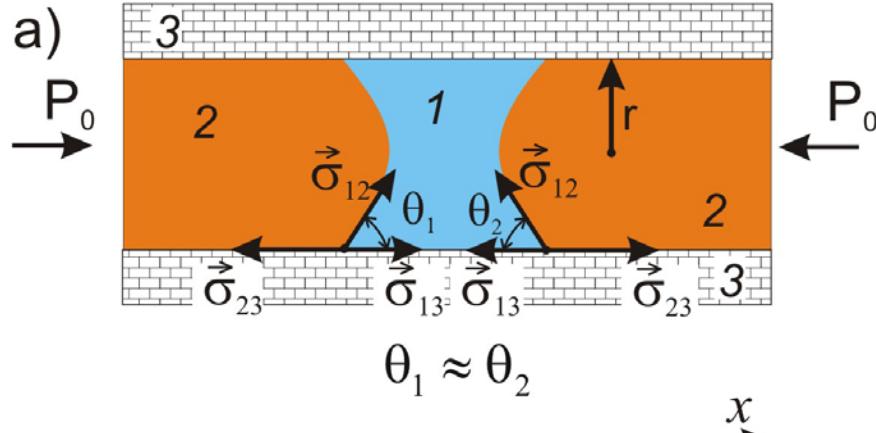
# Inverse and Reversible Switching Gradient Surfaces from Mixed Polyelectrolyte Brushes



Fraction of grafted poly(*tert*-butyl acrylate) vs location for the gradient PBA/P2VP brush.



# EXAMPLES:

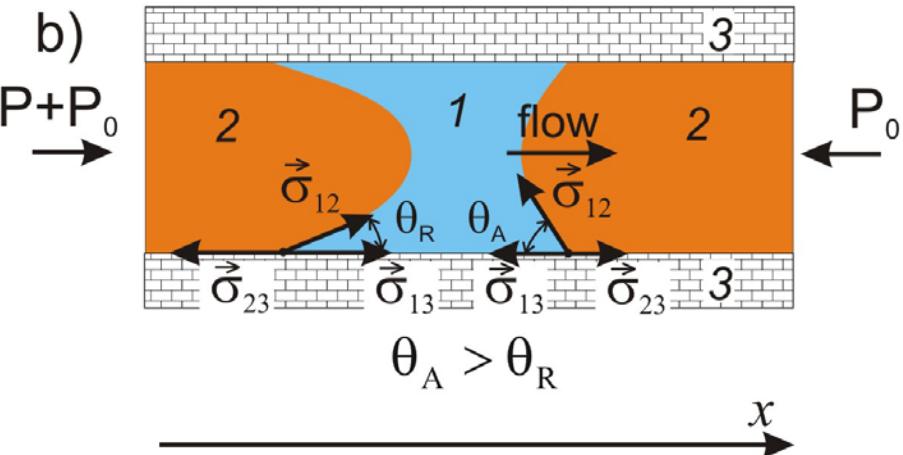


steady state  
no flow

$$\frac{\partial \sigma_{23}}{\partial x} = 0$$

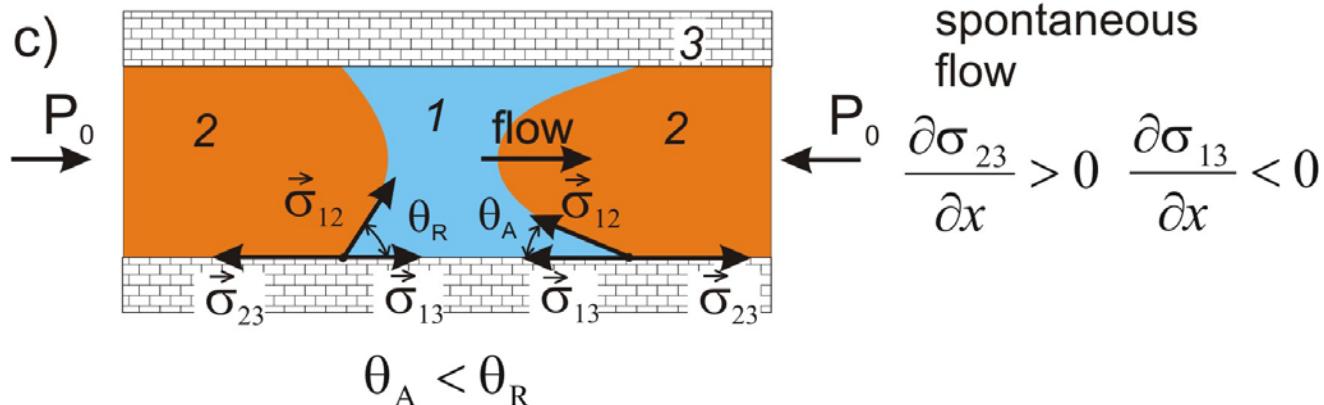
J. C. Jamin, *Phil. Mag.* 1860, 4, 204.

$$P - P_0 = 2 \sigma_{12} (\cos \theta_R - \cos \theta_A) / r,$$



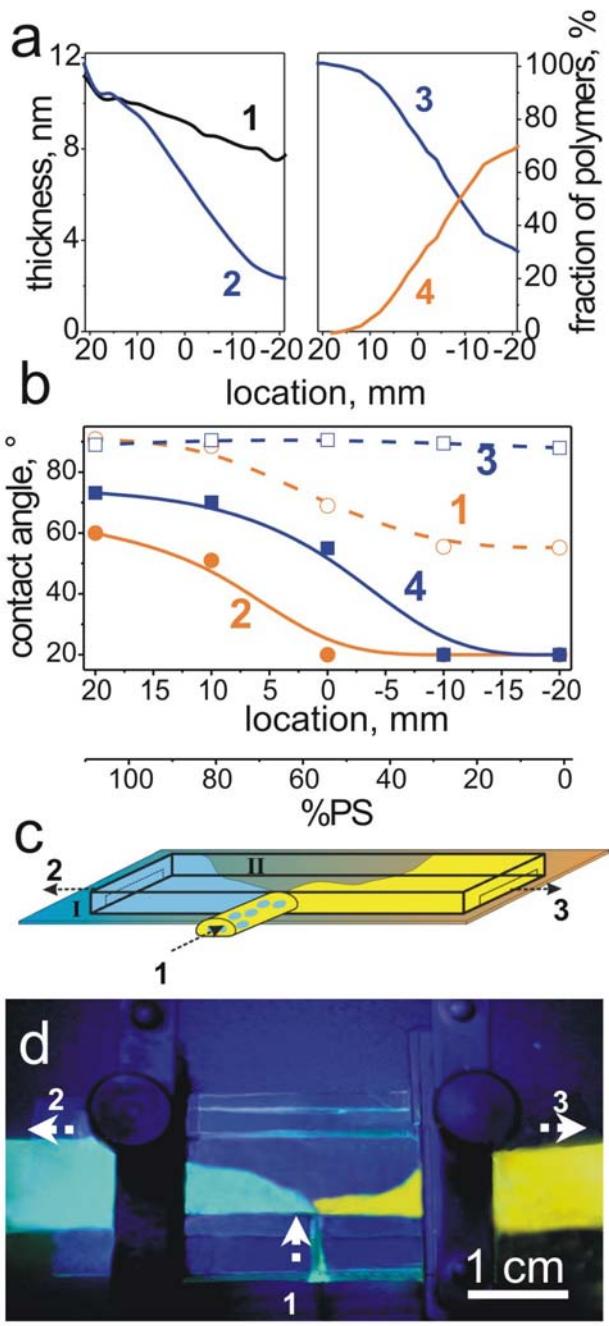
enforced flow

$$\frac{\partial \sigma_{23}}{\partial x} = 0$$

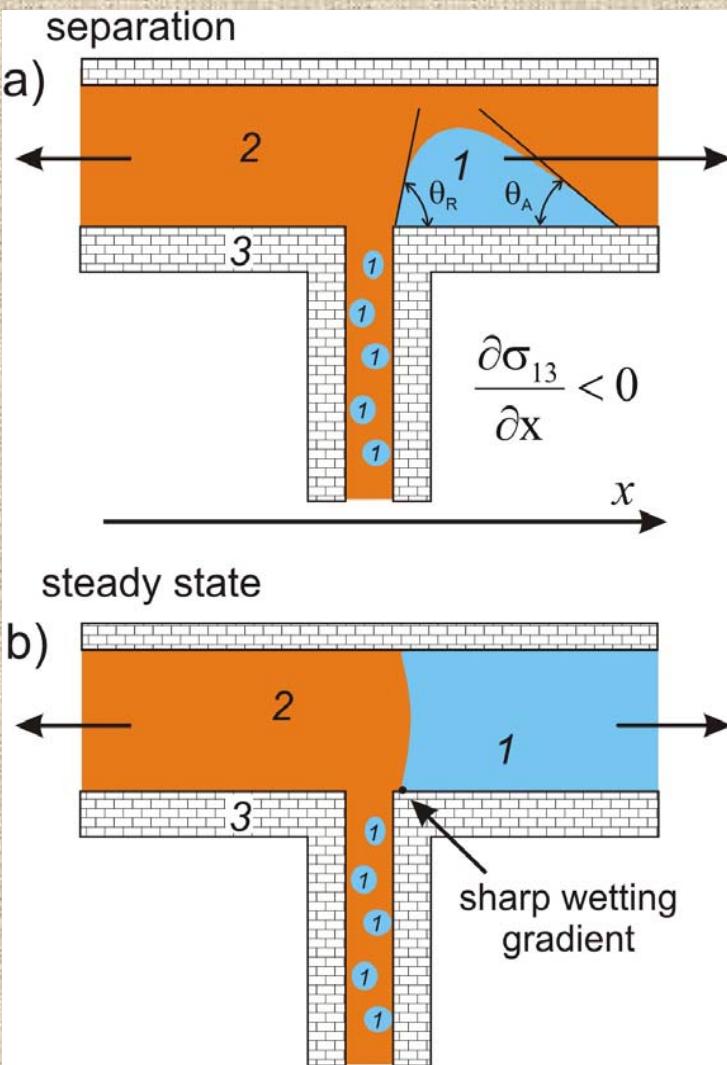


spontaneous  
flow

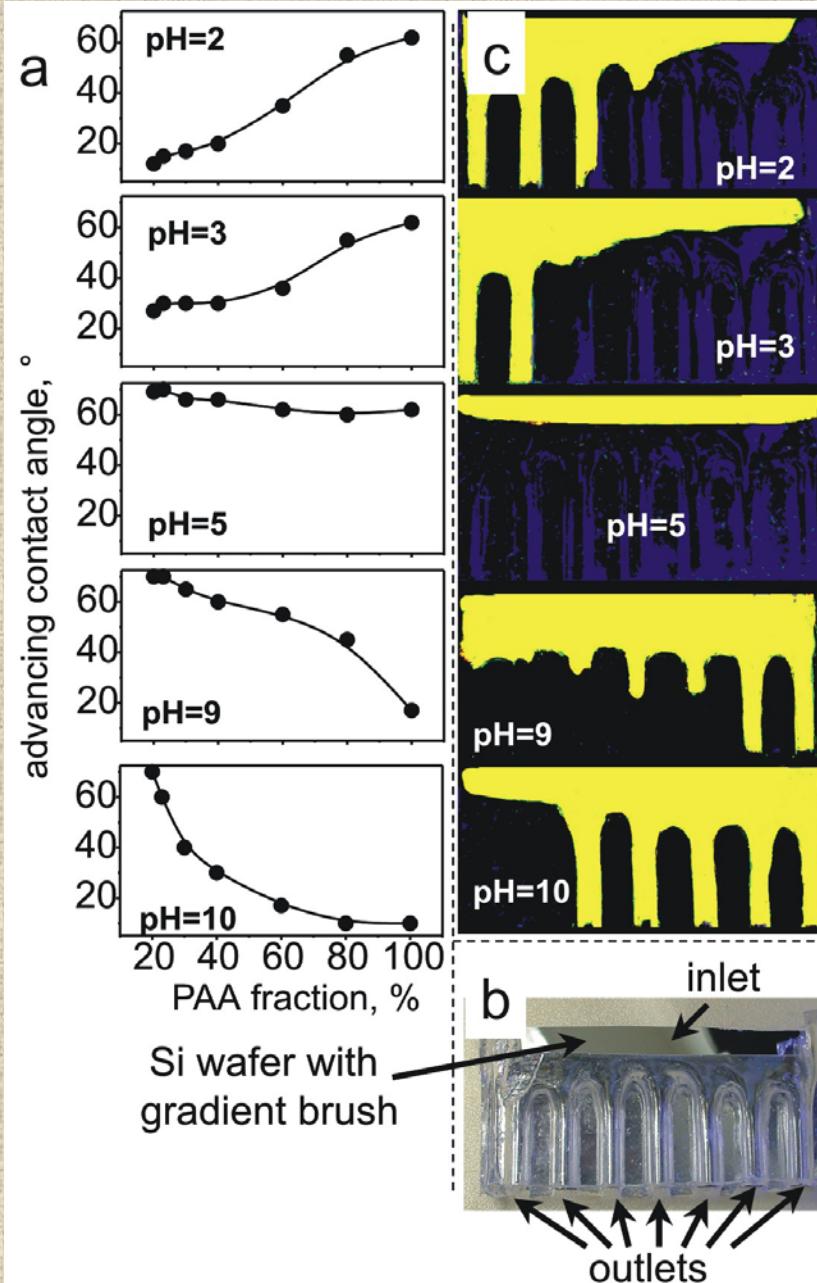
$$\frac{\partial \sigma_{23}}{\partial x} > 0 \quad \frac{\partial \sigma_{13}}{\partial x} < 0$$

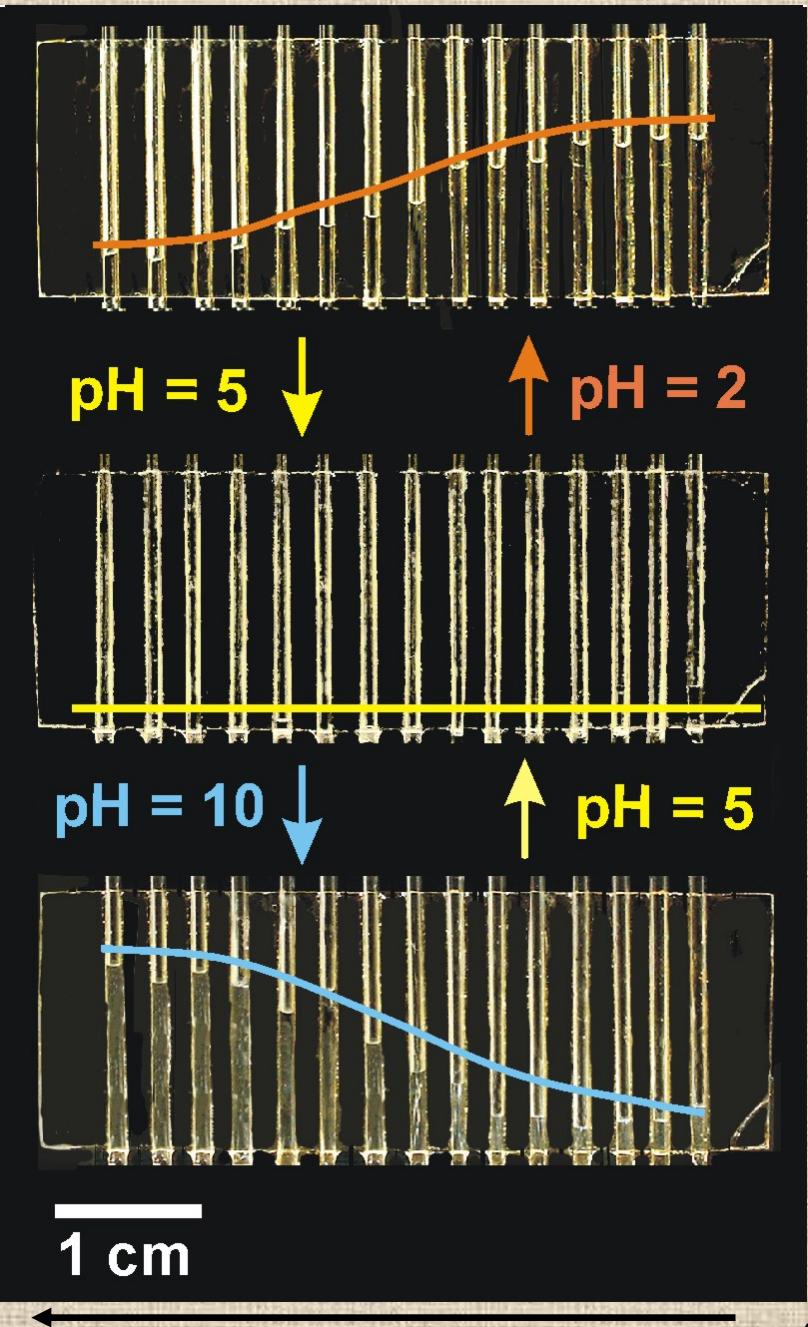


## Separation of emulsions:



# EXAMPLES: Chemical gates

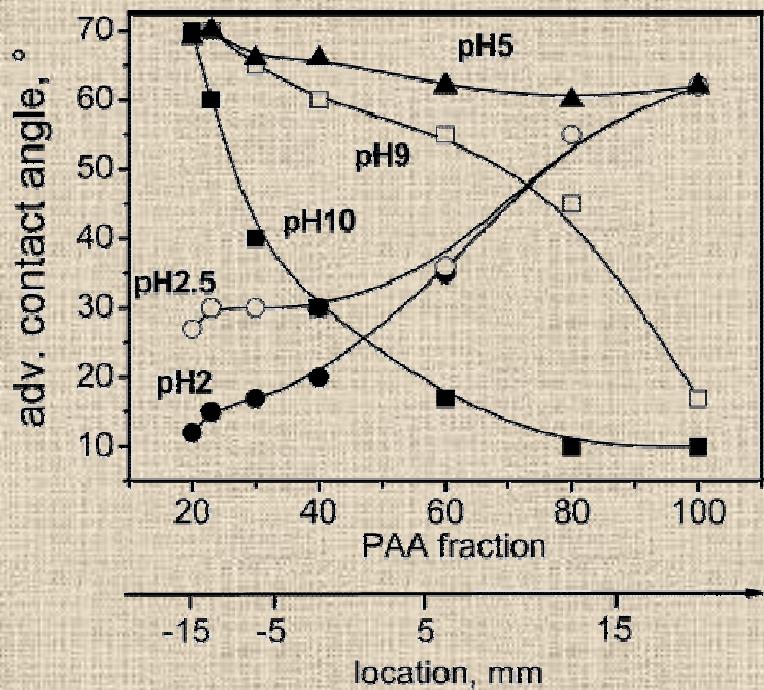




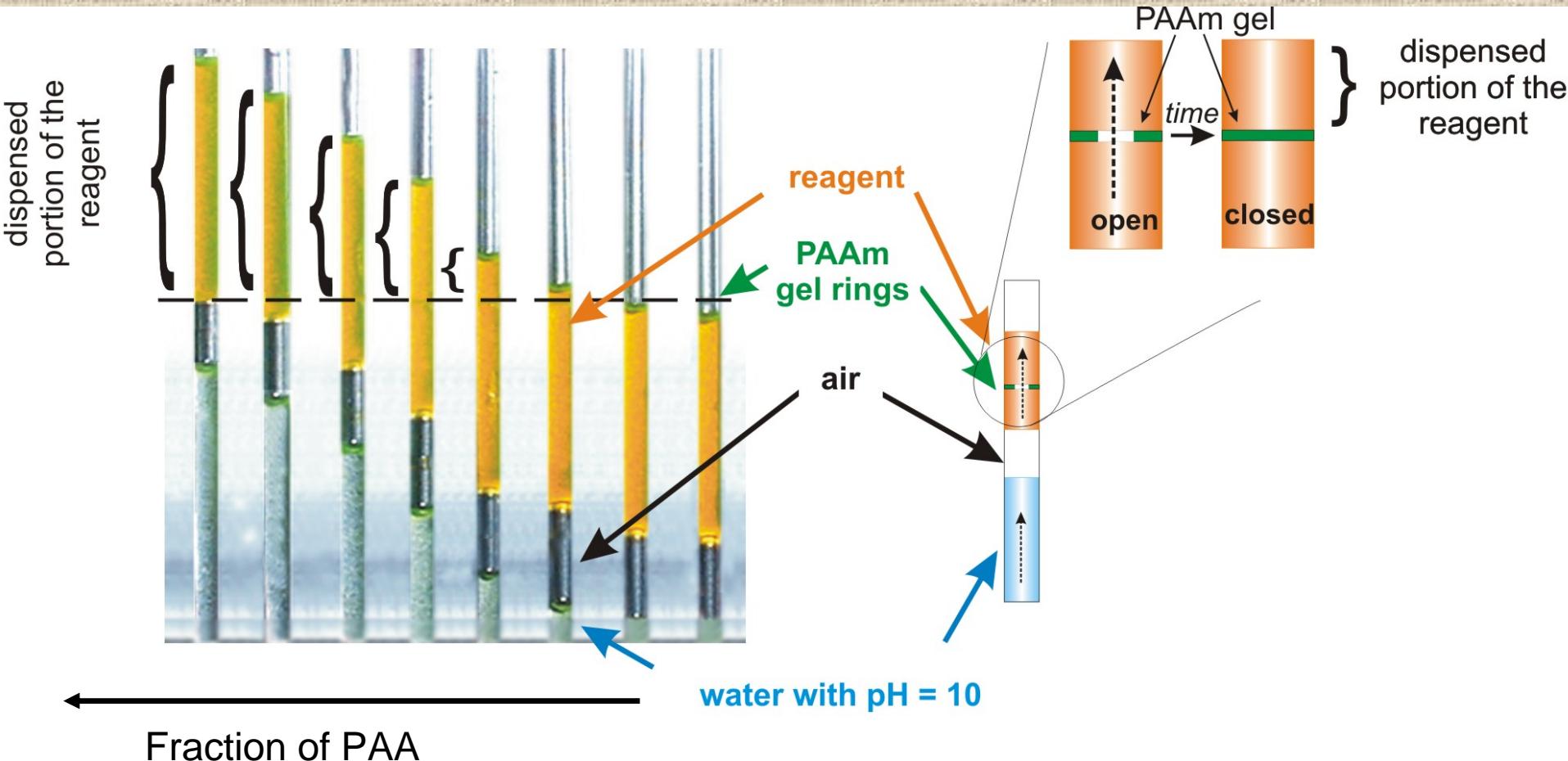
## Capillary Rise

Laplace equation for capillary rise

$$\frac{2\sigma_{12} \cos \theta_A}{r} = p$$

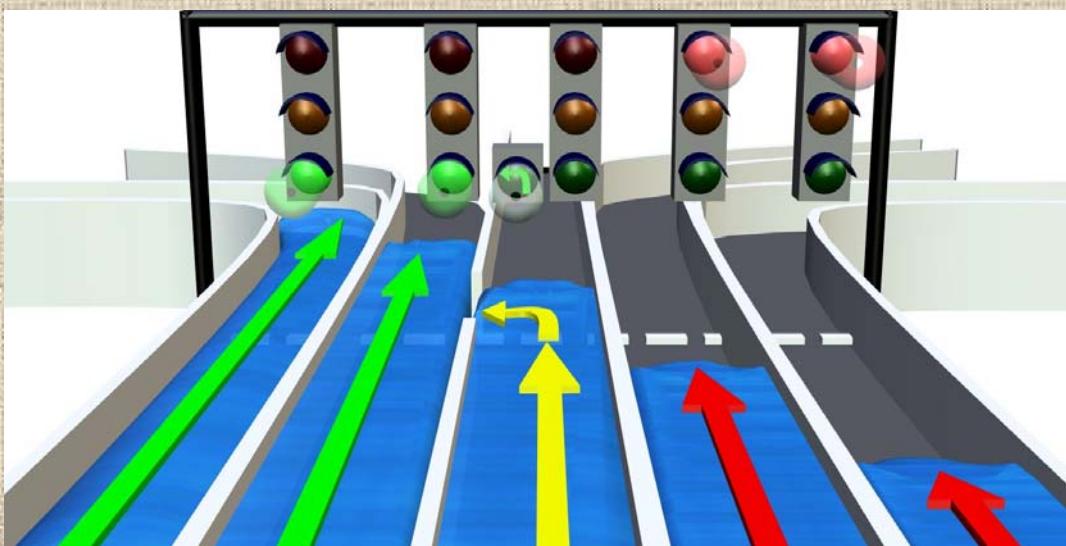


# Microfluidic dispenser



# CONCLUSIONS

- Gradient brushes of complex compositions allow for a precise regulation of interfacial interactions.
- Combinatorial approach is very important for the optimization of the brush composition. Theoretical predictions are limited by simple systems.
- Wetting experiments, adsorption, separation, selection, sensing, and dispensing small volumes of liquids could be used to collect the data about interfacial interactions.



# Acknowledgement:

**Igor Tokarev Clarkson**

**Michael Motornov Clarkson**

**Leonid Ionov, IPF, Germany**

**Nikolai Huobenov, Germany**

**Manfred Stamm, IPF, Dresden, Germany**

**Alex Sidorenko, Bell Labs, NJ**

**Constantinos Tsitsilianis, Univ. Patras, Greece**

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**Financial support:**

**Clarkson University, NSF, NATO, ARO,**

**NYSTAR- CAMP, P&G, Xerox Co.**

**DFG, BMBF, Volkswagen Foundation, AiF**